

Phase space structure and dynamics for the Hamiltonian isokinetic thermostat

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We investigate the phase space structure and dynamics of a Hamiltonian isokinetic thermostat, for which ergodic thermostat trajectories at fixed (zero) energy generate a canonical distribution in configuration space. Model potentials studied consist of a single bistable mode plus transverse harmonic modes. Interpreting the bistable mode as a reaction (isomerization) coordinate, we establish connections with the theory of unimolecular reaction rates, in particular the formulation of isomerization rates in terms of gap times. In the context of molecular reaction rates, the distribution of gap times (or associated lifetimes) for a microcanonical ensemble initiated on the dividing surface is of great dynamical significance; an exponential lifetime distribution is usually taken to be an indicator of “statistical” behavior. Moreover, comparison of the magnitude of the phase space volume swept out by reactive trajectories as they pass through the reactant region with the total phase space volume (classical density of states) for the reactant region provides a necessary condition for ergodic dynamics. We compute gap times, associated lifetime distributions, mean gap times, reactive fluxes, reactive volumes, and total reactant phase space volumes for model thermostat systems with three and four degrees of freedom at three different temperatures. At all three temperatures, the necessary condition for ergodicity is approximately satisfied. At high temperatures a nonexponential lifetime distribution is found, while at low temperatures the lifetime is more nearly exponential. The degree of exponentiality of the lifetime distribution is quantified by computing the information entropy deficit with respect to pure exponential decay. The efficacy of the Hamiltonian isokinetic thermostat is examined by computing coordinate distributions averaged over single long trajectories initiated on the dividing surface. © 2010 American Institute of Physics. [doi:10.1063/1.3455712]

I. INTRODUCTION

Deterministic thermostats are widely used to simulate equilibrium physical systems described by ensembles other than microcanonical [constant energy and volume (E, V)], such as constant temperature-volume (T, V) or temperature-pressure (T, p),^{1–6} or for simulations of nonequilibrium phenomena.^{7–12}

Deterministic thermostats are obtained by augmenting the phase space variables of the physical system of interest with a set of additional variables whose role is to alter the standard Hamiltonian system dynamics in such a way that a suitable invariant measure in the system phase space is preserved.^{1,13,14} For example, in the familiar Nosé–Hoover (NH) thermostat^{13,14} the exact dynamics preserves both an extended energy and a suitable invariant measure, ensuring that, *provided* the extended system dynamics is effectively ergodic on the timescale of the simulation, the physical system will sample its phase space according to the canonical (constant T) measure.

Extended system thermostat dynamics can be either

Hamiltonian^{15–18} or non-Hamiltonian.^{7,19–26} For example, NH dynamics is non-Hamiltonian.¹⁴ (The NH system is however conformally symplectic,^{27,28} meaning that the dynamics can be rendered Hamiltonian by a coordinate-dependent scaling of the vector field.²⁷ This is not true for other non-Hamiltonian thermostats such as NH chains²⁹ or Bulgac–Kusnezov global demons.³⁰)

There has been considerable interest in the formulation of Hamiltonian deterministic thermostats such as the Nosé–Poincaré system.¹⁸ In this approach, the extended Hamiltonian for the physical system plus thermostat variables incorporates a coordinate-dependent time scaling of Poincaré–Sundman type.^{31,32} Restricting the dynamics to a fixed value (zero) of the extended Hamiltonian results in the system variables sampling their phase space according to (for example) the canonical density¹⁸ (again, subject to the assumption of ergodicity). An important motivation for the introduction of Hamiltonian thermostats is the possibility of using symplectic integration algorithms to integrate trajectories.^{4,33,34}

As already indicated, a fundamental question concerning deterministic thermostats has to do with the effective ergodicity of the dynamics on the timescale of the simulation. If the dynamics is not effectively ergodic, then trajectory simulations will not generate the correct invariant measure.^{35,36} It

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has long been recognized, for example, that the dynamical system consisting of a single harmonic oscillator degree of freedom (DOF) coupled to the NH thermostat variable is not ergodic¹⁴ (see also Refs. 37–39); in fact, for typical parameter values the system phase space exhibits extensive persistence of invariant tori (quasiperiodic motion).^{40,41} A large amount of effort has been expended in attempts to design thermostats exhibiting dynamics more ergodic than the basic NH system.^{4,5,29,42} (For a careful discussion of the question of ergodicity for “typical” interatomic potentials, see Ref. 43; for related fundamental questions in molecular dynamics, see also Ref. 44.)

The question of ergodicity in thermostats is conceptually closely related to the problem of statistical versus nonstatistical behavior in the (classical) theory of unimolecular reaction rates.^{45–48} Broadly speaking, in this case one would like to know whether a molecule will behave according to a statistical model such as RRKM theory, or whether it will exhibit significant deviations from such a theory, ascribable to nonstatistical dynamics.^{49–54} There are clearly direct parallels between “nonstatistical” behavior in molecules, which can arise from a number of dynamical effects, and the failure of ergodicity in deterministic thermostats.

In recent years there have been significant theoretical and computational advances in the application of dynamical systems theory^{55–57} to study reaction dynamics and phase space structure in multimode models of molecular systems, and to probe the dynamical origins of nonstatistical behavior.^{58–66} The fundamental chemical concept of the *transition state*, defined as a surface of no return in phase space, has been successfully and rigorously generalized from the well-established two DOF case⁶⁷ to systems with $N \geq 3$ DOF.^{61–66} Moreover, dynamical indicators exist (determination of reactive phase space volume, behavior of the reactive flux) to diagnose nonstatistical behavior (see, for example, Refs. 51, 54, and 68).

Nevertheless, relatively little work has been done applying the powerful techniques from dynamical system theory^{55–57} to study the phase space structure and dynamics of deterministic thermostats. Hoover and co-workers⁶⁹ have investigated the fractal nature of various phase space structures for equilibrium and nonequilibrium^{70–72} versions of the NH thermostat. Using a Hamiltonian formulation of the isokinetic thermostat¹⁵ (cf. Sec. II of this paper), Morriss and Dettmann² have mapped the dynamics of the isokinetically thermostatted Lorentz gas onto the geodesic motion of a free particle on a particular Riemannian manifold. Leimkuhler and Sweet⁴² have used Poincaré surfaces of section and dynamical frequency analysis in their analysis of optimal coupling constants for NH and related thermostats, while Legoll and co-workers^{40,41} have applied KAM theory to rigorously prove the existence of invariant tori in the NH thermostatted harmonic oscillator. D’Alessandro, Tenenbaum, and Amadei⁷³ have computed Lyapunov exponents for a thermostatted united-atom model of the butane molecule; both NH and isokinetic Gaussian thermostats were considered.

Thermostatted systems present several difficulties for detailed studies from a phase space perspective. First, such systems usually have high dimensionality; for example, a NH

chain thermostatted system has at least three DOF. For such a system the Poincaré surface of section analysis, standard for two DOF, does not afford the advantage of dimensional reduction and global visualization since the surface of section is four dimensional (4D) for three DOF. Second, there is a lack of readily computable diagnostics that can be used to establish ergodicity,³⁶ other than comparison of coordinate distributions obtained using the given thermostat with those obtained using other methods.

In the present paper we analyze the Hamiltonian isokinetic thermostat² from the perspective of reaction rate theory. Although not as widely used as the NH thermostat and its many variants, the non-Hamiltonian version of the isokinetic thermostat has been developed and applied to several problems of chemical interest by Minary *et al.*^{74–77} In this thermostat, the particle momenta are subject to a nonholonomic constraint that keeps the kinetic energy constant. The resulting dynamics generates a canonical (constant temperature) distribution in configuration space.² In this work we investigate a slightly modified version of the Hamiltonian isokinetic thermostat given by Litniewski⁷⁸ and Morishita.⁷⁹

The structure of the present paper is as follows. Section II reviews the Hamiltonian formulation of the isokinetic thermostat. The non-Hamiltonian equations of motion for a Hamiltonian system subject to the isokinetic constraint correspond to Hamiltonian dynamics at zero energy under an effective Hamiltonian whose potential is obtained from the physical potential by exponentiation. The model Hamiltonians analyzed in the present paper are introduced in Sec. III. In these systems, the physical potential describes n uncoupled oscillators, $n-1$ harmonic modes together with a bistable thermalizing^{74,75} or isomerization coordinate. For these Hamiltonians the physical potential exhibits a saddle of index one, as for the case of a bistable reaction profile coupled to one or more transverse confining modes. In Sec. IV we briefly review earlier results⁸⁰ establishing that the isokinetic Hamiltonian dynamical system corresponding to the physical Hamiltonian exhibits a normally hyperbolic invariant manifold (NHIM) associated with the saddle,⁵⁷ at least for a limited range of energies above that of the saddle. The phase space formulation of unimolecular reaction rate theory in terms of the gap time^{54,81–83} and related concepts is discussed in Sec. V. The classical spectral theorem^{84–89} provides a relation between the distribution of gap times and the phase space volume occupied by reactive phase points; unless the measure of the region swept out by isomerizing trajectories equals that of the energy shell, the system cannot be ergodic. Ergodicity is a necessary condition for a thermostat to generate the correct phase space distribution; our use of a necessary condition for ergodicity derived via the spectral theorem establishes a significant connection between the concepts of reaction rate theory and the properties of the Hamiltonian isokinetic thermostat. Note, however, that kinetic parameters derived from the gap distributions we compute for our model thermostats have no direct connection with rate processes occurring in the physical system of interest; the analysis of lifetimes, etc., is simply a means to characterize thermostat dynamics. Numerical results on three and four DOF Hamiltonian isokinetic thermostats are reported in

Sec. VI. Section VII concludes. Analytical results for the density of states for exponentiated harmonic potential are given in Appendix A, while details of the phase space sampling procedures used in our numerical work appear in Appendix B.

A brief discussion of the theoretical framework developed here was given in Ref. 80.

II. THE HAMILTONIAN ISOKINETIC THERMOSTAT

The role of the isokinetic thermostat is to dynamically constrain the kinetic energy of a simulated system to have a constant value proportional to $k_B T$, where T is the desired temperature and k_B is Boltzmann's constant. Non-Hamiltonian equations of motion for the isokinetic thermostat have been obtained by applying Gauss' principle of least constraint,^{2,7,8} which is the appropriate dynamical principle for nonholonomically constrained systems.^{7,90}

Provided the underlying dynamics is effectively ergodic on the timescale of the simulation, a trajectory of the non-Hamiltonian isokinetic thermostat samples configuration space according to the invariant measure associated with the equilibrium canonical ensemble.^{2,91} Algorithms for the non-Hamiltonian isokinetic thermostat have been developed and applied to a variety of systems by Minary, Martyna, and Tuckerman.^{74,75} As well as model systems, the isokinetic method has been applied to study mechanisms of organic reactions on semiconductor surfaces.^{76,77}

A Hamiltonian formulation of the isokinetic thermostat has been given by Dettmann and Morriss.^{2,15} The Hamiltonian used in the Dettmann–Morriss approach incorporates a coordinate-dependent scaling of time,^{32,92} and the physically relevant dynamics is restricted to the zero energy hypersurface. A noncanonical transformation of variables then leads to a set of dynamical equations equivalent to the non-Hamiltonian version. An important motivation for the development of Hamiltonian versions of the isokinetic thermostat is the possibility of using symplectic integration algorithms to ensure qualitatively correct behavior of integrated trajectories over long times.^{33,34}

In this section we briefly review the Hamiltonian formulation of the isokinetic thermostat (for more detailed discussion, see Ref. 80). As our focus here is on systems with $n \geq 3$ DOF, we discuss a Hamiltonian version of the isokinetic thermostat that generates the invariant measure associated with the configurational canonical ensemble, while at the same time allowing use of the simplest Verlet-type⁹³ symplectic integration algorithm.^{78,79}

A. Hamiltonian isokinetic thermostat

The physical Hamiltonian of interest is assumed to have the standard form

$$H(q, p) = \frac{1}{2} p^2 + \Phi(q), \quad (1)$$

where $(q, p) \in \mathbb{R}^n \times \mathbb{R}^n$, $\Phi(q)$ is the potential energy, and we set all masses equal to unity for simplicity. Corresponding physical Hamiltonian equations of motion are

$$\dot{q} = + \frac{\partial H}{\partial p}, \quad (2a)$$

$$\dot{p} = - \frac{\partial H}{\partial q}. \quad (2b)$$

For $n \geq 3$ DOF, we define a Hamiltonian $\mathcal{H} = \mathcal{H}(\pi, q)$ as^{78,79}

$$\mathcal{H} = \frac{1}{2} \pi^2 - \frac{\nu}{2\bar{\beta}} e^{-2\bar{\beta}\Phi}, \quad (3)$$

where $\pi \in \mathbb{R}^n$ and $\bar{\beta}$ and ν are parameters to be determined. The associated Hamiltonian equations of motion are

$$q' = + \frac{\partial \mathcal{H}}{\partial \pi} = \pi, \quad (4a)$$

$$\pi' = - \frac{\partial \mathcal{H}}{\partial q} = - \nu \Phi_q e^{-2\bar{\beta}\Phi}, \quad (4b)$$

where the time derivative is denoted by a prime (') to indicate that the derivative is actually taken with respect to a suitably scaled time variable (see Refs. 2 and 80). As the kinetic energy term in Eq. (3) is coordinate-independent, a basic Verlet-type symplectic integrator can be used to integrate Hamilton's equations for \mathcal{H} .^{4,33,34,93} (Higher-order symplectic algorithms can of course also be used.^{4,34})

Making a *noncanonical* change of coordinates from variable (π, q) to physical variables (p, q) , $(\pi, q) \mapsto (p, q)$, where

$$\pi = e^{-\bar{\beta}\Phi} p, \quad (5)$$

the Hamiltonian function \mathcal{H} becomes

$$\mathcal{H} = e^{-2\bar{\beta}\Phi} \frac{1}{2} \left[p^2 - \frac{\nu}{\bar{\beta}} \right]. \quad (6)$$

Setting $\mathcal{H} = 0$ then automatically enforces an *isokinetic constraint* in terms of the momentum variables p (Ref. 2),

$$\frac{p^2}{2} = \frac{\nu}{2\bar{\beta}}. \quad (7)$$

For trajectories run at $\mathcal{H} = 0$, and only for this value, we have the invariant volume element²

$$dV = d^n \pi d^n q \delta(\mathcal{H}), \quad (8a)$$

$$= d^n p d^n q e^{-n\bar{\beta}\Phi} \delta(\mathcal{H}), \quad (8b)$$

$$= d^n p d^n q e^{-n\bar{\beta}\Phi} \delta \left[\frac{1}{2} e^{-2\bar{\beta}\Phi} \left(p^2 - \frac{\nu}{\bar{\beta}} \right) \right], \quad (8c)$$

$$= d^n p d^n q 2 e^{-(n-2)\bar{\beta}\Phi} \delta \left[p^2 - \frac{\nu}{\bar{\beta}} \right]. \quad (8d)$$

We therefore set $(\beta \equiv 1/k_B T)$ as usual

$$\bar{\beta} = \frac{1}{(n-2)k_B T} = \frac{\beta}{(n-2)} \quad (9)$$

to ensure that the invariant measure is proportional to the canonical density (recall $n > 2$),

$$\rho(q) \propto \exp[-\Phi(q)/k_B T]. \quad (10)$$

With

$$\nu = \frac{n}{(n-2)}, \quad (11)$$

the kinetic energy

$$\frac{p^2}{2} = \frac{n}{2} k_B T, \quad (12)$$

as required for an n DOF system at temperature T . Nevertheless, the important aspect of the dynamics for computing configurational averages is the invariant measure, not the magnitude of the constrained kinetic energy. The quantity ν can then be treated as a free parameter, which can be used to move the potential saddle closer to energy $\mathcal{E}=0$ (cf. Sec. IV). In fact, we shall take $\nu=1$ in our calculations.

For $n \geq 3$ DOF, and provided the dynamics is ergodic on the $\mathcal{H}=0$ energy surface, Hamiltonian dynamics (4) will therefore yield the canonical measure in configuration space. Since the aim is to obtain a canonical distribution, the Hamiltonian dynamics derived using Eq. (3) is in this respect equivalent in principle to the original isokinetic thermostat.²

For completeness, we note that, in order to treat the $n=2$ DOF case, it is necessary to use a different Hamiltonian \mathcal{K} (cf. Refs. 2 and 80) with

$$\mathcal{K} = \frac{1}{2} e^{\bar{\beta}\Phi} \pi^2 - \frac{\nu}{2\bar{\beta}} e^{-\bar{\beta}\Phi}. \quad (13)$$

To obtain the correct canonical measure case we must then take ($n \geq 2$)

$$\bar{\beta} = \frac{1}{(n-1)k_B T} = \frac{\beta}{(n-1)} \quad (14)$$

and

$$\nu = \frac{n}{(n-1)}. \quad (15)$$

III. MODEL HAMILTONIANS

In this section we introduce the model Hamiltonians studied in the remainder of the paper. We define systems with three and four DOF suitable for studying the Hamiltonian isokinetic thermostat on the zero energy surface $\mathcal{H}=0$.

The phase space structure of these systems is discussed in Sec. IV, while numerical computations of gap times, reactive volumes, and coordinate distributions obtained using the Hamiltonian isokinetic thermostat are described in Sec. VI.

The systems considered here consist of n uncoupled oscillators: $n-1$ harmonic modes plus a single bistable mode. Although the n modes are uncoupled in the physical Hamiltonian $H(p, q)$, exponentiation of the potential $\Phi(q)$ intro-

duces intermode coupling in the isokinetic thermostat Hamiltonian \mathcal{H} . The bistable mode can be interpreted either as a thermalizing DOF (following Minary, Martyna, and Tuckerman^{74,75}), or as a reactive DOF associated with an isomerization process. Adopting the latter perspective, we can apply concepts and methods recently developed for understanding reaction dynamics (in particular, transition state theory) in phase space^{54,61-66} to investigate the problem of thermostat dynamics.

A. Double well potential

The double well potential for the thermalizing mode is taken to be a temperature-independent quartic having the following form:

$$\chi(y) = \frac{1}{2}(y^4 - \alpha y^2). \quad (16)$$

(Note that the Minary, Martyna, and Tuckerman^{74,75} version of the double well potential is effectively temperature dependent.) The potential (16) has stationary points at $y=0$ (a maximum) and $y = \pm \sqrt{\alpha/2}$ (minima), with corresponding values 0 and $-\alpha^2/8$. Expanding $\chi(y)$ about the minima $y = \pm \sqrt{\alpha/2}$ we find that the effective frequency for harmonic motion in the vicinity of the minima is

$$\bar{\omega} = \sqrt{2\alpha}. \quad (17)$$

B. Model Hamiltonians

1. Three degrees of freedom

The three DOF potential corresponds to a separable bistable (y) plus harmonic bath modes (x_1, x_2),

$$\Phi(x_1, x_2, y) = \frac{m\omega_1^2 x_1^2}{2} + \frac{m\omega_2^2 x_2^2}{2} + \frac{1}{2}(y^4 - \alpha y^2). \quad (18)$$

Setting $m=1$ and choosing potential parameters $\omega_1=1$, $\omega_2 = \sqrt{2}$, and $\alpha=2$, we have

$$\Phi(x_1, x_2, y) = \frac{x_1^2}{2} + x_2^2 + \frac{y^4}{2} - y^2. \quad (19)$$

The isokinetic thermostat Hamiltonian \mathcal{H} is then

$$\begin{aligned} \mathcal{H}(x_1, x_2, y, \pi_{x_1}, \pi_{x_2}, \pi_y) \\ = \frac{\pi_{x_1}^2}{2} + \frac{\pi_{x_2}^2}{2} + \frac{\pi_y^2}{2} - \frac{\nu}{2\beta} \\ \times \exp\left[-2\beta\left\{\frac{x_1^2}{2} + x_2^2 + \frac{y^4}{2} - y^2\right\}\right]. \end{aligned} \quad (20)$$

The origin is an equilibrium point of saddle-center-center type with energy $\mathcal{H}(0, 0, 0, 0, 0, 0) = -\nu/2\beta$, and ν and β are parameters that we can vary. As mentioned above, the value of the parameter ν determines the (constant) value of the physical kinetic energy $p^2/2$. It also determines the energy of the saddle point with respect to the zero of energy, $\mathcal{H}=0$. To obtain the correct canonical invariant density, it is only necessary that p^2 be constant; the parameter ν is therefore effectively a free parameter in addition to the temperature T .

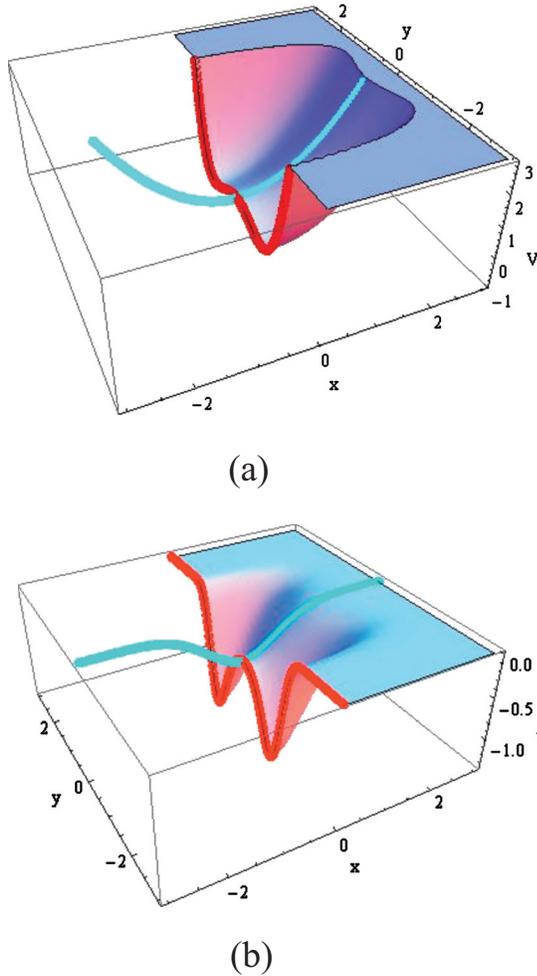


FIG. 1. (a) Section ($x_2=0$) through the physical potential $\Phi(x_1, x_2, y)$, Eq. (19), $\omega_1=1$, $\omega_2=\sqrt{2}$, $\alpha=2$. (b) Section ($x_2=0$) through the exponentiated potential, $-\frac{1}{2}\exp[-2\beta\Phi]$, $\beta=1$.

For the numerical computations to be discussed below we take $\alpha=2$, $\nu=1$. Plots of the $x_2=0$ slice through the physical potential Φ for the three DOF system and the corresponding exponentiated potential ($\beta=1$) are shown in Fig. 1.

2. Four degrees of freedom

The four DOF potential corresponds to a separable bistable (y) plus harmonic bath modes (x_1, x_2, x_3),

$$\Phi(x_1, x_2, x_3, y) = \frac{m\omega_1^2 x_1^2}{2} + \frac{m\omega_2^2 x_2^2}{2} + \frac{m\omega_3^2 x_3^2}{2} + \frac{1}{2}(y^4 - \alpha y^2). \quad (21)$$

Setting $m=1$, $\omega_1=1$, $\omega_2=\sqrt{2}$, $\omega_3=\sqrt{3}$, and $\alpha=2$, we have

$$\Phi(x_1, x_2, x_3, y) = \frac{x_1^2}{2} + x_2^2 + \frac{3x_3^2}{2} + \frac{y^4}{2} - y^2. \quad (22)$$

From Eq. (9), the parameter $\bar{\beta}=\beta/2$ for $n=4$ DOF, so that the isokinetic thermostat Hamiltonian \mathcal{H} is

$$\begin{aligned} \mathcal{H}(x_1, x_2, x_3, y, \pi_{x_1}, \pi_{x_2}, \pi_{x_3}, \pi_y) \\ = \frac{\pi_{x_1}^2}{2} + \frac{\pi_{x_2}^2}{2} + \frac{\pi_{x_3}^2}{2} + \frac{\pi_y^2}{2} - \frac{\nu}{\beta} \\ \times \exp\left[-\beta\left\{\frac{x_1^2}{2} + x_2^2 + \frac{3x_3^2}{2} + \frac{y^4}{2} - y^2\right\}\right]. \end{aligned} \quad (23)$$

The origin is an equilibrium point of saddle-center-center-center type with energy $\mathcal{H}(0,0,0,0,0,0,0,0)=-\nu/\beta$, and ν and β are parameters that we can vary. We set the parameter $\nu=1$.

In Sec. VI we present numerical results for both the four DOF and three DOF systems corresponding to three values of the temperature: $\beta=1, 3$, and 5 .

IV. MICROCANONICAL PHASE SPACE STRUCTURE: HAMILTONIAN AND NON-HAMILTONIAN ISOKINETIC THERMOSTAT

In this section, we briefly discuss the phase space structure in the vicinity of the saddle-center equilibrium points for Hamiltonians (1) and (3).

In previous work, we have established the following:⁸⁰

- If the physical Hamiltonian system defined by Eq. (1) has an equilibrium point of saddle-center-...-center stability type⁵⁷ at the origin then the Hamiltonian system defined by an extended Hamiltonian of the form (13) corresponding to the isokinetic thermostat has an equilibrium point at the origin of saddle-center-...-center stability type.
- The energy of the equilibrium is such that on the zero energy surface of the extended Hamiltonian \mathcal{K} the phase space structures present in the physical Hamiltonian also exist in the extended Hamiltonian phase space.
- The phase space structures on the zero energy surface corresponding to the Hamiltonian isokinetic thermostat map to phase space structures in the non-Hamiltonian thermostatted system obtained by transforming the Hamiltonian equations of motion for evolution (π, q) under \mathcal{K} to equations of motion for noncanonical variables (p, q) under the constraint of zero energy.

Although these results were previously established for the Hamiltonian \mathcal{K} , Eq. (13), they also hold for Hamiltonian \mathcal{H} , Eq. (3), as we describe below.

More precisely, we have the following: Assume that the physical Hamiltonian system (2) has an equilibrium point at $(q, p)=(q^*, p^*)=(0,0)$. The energy of this equilibrium point is $H(0,0)=\Phi(0)$. The stability of the equilibrium point is determined by the eigenvalues of the derivative of the Hamiltonian vector field (or Hessian of the Hamiltonian function) evaluated at the equilibrium point. This is given by the $2n \times 2n$ matrix

$$\text{Hess}_{\text{sys}} = \begin{pmatrix} 0_{n \times n} & \text{id}_{n \times n} \\ -\Phi_{qq}(0) & 0_{n \times n} \end{pmatrix}, \quad (24)$$

where $0_{n \times n}$ denotes the $n \times n$ matrix of zeros and $\text{id}_{n \times n}$ denotes the $n \times n$ identity matrix. We require the equilibrium

point to be of saddle-center- \cdots -center stability type. This means that the $2n \times 2n$ matrix Hess_{sys} has eigenvalues $\pm\lambda, \pm i\omega_i, i=2, \dots, n$ where λ and ω_i are real.

Eigenvalues γ of Hess_{sys} are obtained by solving the characteristic equation $\det(\text{Hess}_{\text{sys}} - \gamma \text{id}_{2n \times 2n}) = 0$. The block structure of the $2n \times 2n$ matrix Hess_{sys} implies that (cf. Theorem 3 of Ref. 94)

$$\det(\text{Hess}_{\text{sys}} - \gamma \text{id}_{2n \times 2n}) = \det(\Phi_{qq}(0) + \gamma^2 \text{id}_{n \times n}) = 0, \quad (25)$$

so that the $2n$ eigenvalues γ are given in terms of σ , the eigenvalues of the $n \times n$ Hessian matrix $\Phi_{qq}(0)$ associated with the potential, as follows:

$$\gamma_k, \gamma_{k+n} = \pm \sqrt{-\sigma_k}, \quad k = 1, \dots, n. \quad (26)$$

Therefore, if $\Phi(q)$ has a rank-one saddle at $q=0$, so that one eigenvalue is strictly negative and the rest are strictly positive, then $(q, p) = (0, 0)$ is a saddle-center- \cdots -center type equilibrium point for Eq. (2) as described above.

Next, we consider Hamilton's equations associated with the extended Hamiltonian (3), Eq. (4), corresponding to the isokinetic thermostat. It is easy to verify that $(q, \pi) = (0, 0)$ is an equilibrium point for Eq. (4) with energy $\mathcal{H}(0, 0) = (-\nu/2\bar{\beta})e^{-2\bar{\beta}\Phi(0)}$.

Proceeding as above, we linearize Eq. (4) about $(q, \pi) = (0, 0)$ and compute the eigenvalues of the matrix associated with the linearization. These are given by

$$\tilde{\gamma}_k, \tilde{\gamma}_{k+n} = \pm \sqrt{\nu e^{-\bar{\beta}\Phi(0)}} \sqrt{-\sigma_k}, \quad k = 1, \dots, n. \quad (27)$$

In other words, the eigenvalues of the linearization of Eq. (4) about $(q, \pi) = (0, 0)$ correspond to the eigenvalues of the matrix associated with the linearization of Eq. (2) about $(q, p) = (0, 0)$, but with each eigenvalue multiplied by the positive constant $\sqrt{\nu e^{-\bar{\beta}\Phi(0)}}$. Hence, it follows that if the potential of the physical Hamiltonian $\Phi(q)$ has a rank-one saddle at $q=0$, so that one eigenvalue is strictly negative and the rest are strictly positive, then $(q, \pi) = (0, 0)$ is a saddle-center- \cdots -center type equilibrium point for Eq. (4). Moreover, if the purely imaginary eigenvalues in Eq. (26) satisfy a nonresonance condition, then the purely imaginary eigenvalues in Eq. (27) satisfy the *same* nonresonance condition.⁸⁰

The equilibrium point $(q, \pi) = (0, 0)$ has energy $\mathcal{H}(0, 0) = (-\nu/2\bar{\beta})e^{-2\bar{\beta}\Phi(0)}$. However, we are only interested in the dynamics on the $\mathcal{H}=0$ energy surface. All of the phase space structure discussed above exists for a certain range of energies above that of the saddle-center- \cdots -center, and will do so on the $\mathcal{H}=0$ surface if $(-\nu/2\bar{\beta})e^{-2\bar{\beta}\Phi(0)}$ is close enough to zero.

Putting together the results above, we have the following.⁸⁰

Suppose the physical Hamiltonian system (2) has an equilibrium point of saddle-center- \cdots -center stability type at the origin. Then the Hamiltonian system (4) corresponding to the isokinetic thermostat for $n \geq 3$ DOF also has an equilibrium point of saddle-center- \cdots -center stability type at the origin. Moreover, the energy of this equilibrium point can be chosen so that on the energy surface corresponding to $\mathcal{H}=0$ there exists a NHIM,⁵⁷ with associated stable and un-

stable manifolds, a “dividing surface” (DS) of no-return and minimal flux, and a foliation of the reaction region by n -dimensional invariant Lagrangian submanifolds.⁶⁶

Following the general arguments outlined in Ref. 80 it is straightforward to show that the phase space structure of Eq. (4) on $\mathcal{H}=0$ exists in the non-Hamiltonian thermostatted system in the original physical variables. These general results allow us to conclude that under the noncanonical transformation of variables $(\pi, q) \mapsto (p, q)$:

- The $2n-1$ dimensional invariant energy surface $\mathcal{H}=0$ corresponding to the Hamiltonian isokinetic thermostat (4) maps to a $2n-1$ dimensional invariant manifold for the non-Hamiltonian thermostatted system.
- The $2n-3$ dimensional NHIM, its $2n-2$ dimensional stable and unstable manifolds, the n -dimensional invariant Lagrangian submanifolds, and the $2n-2$ dimensional DS map to a $2n-3$ dimensional NHIM, its $2n-2$ dimensional stable and unstable manifolds, n -dimensional invariant submanifolds, and the $2n-2$ dimensional DS in the $2n-1$ dimensional invariant manifold for the non-Hamiltonian thermostatted system.

V. PHASE SPACE GEOMETRICAL STRUCTURES, UNIMOLECULAR REACTION RATES, AND THERMOSTAT DYNAMICS: GAP TIMES AND REACTIVE VOLUMES

Our analysis of thermostat dynamics will be carried out in phase space, and our approach to probing the dynamics, and especially the question of ergodicity, will be based on the general formulation of unimolecular reaction rates based upon that originally given by Thiele.⁸² In their general form the rate expressions derived by Thiele explicitly invoke the existence of a *phase space* DS separating reactants and products; such surfaces, discussed by Wigner⁹⁵ (see also Refs. 96 and 97), have only recently become amenable to direct computation via the use of normal form approaches.^{61-63,66,88,89,98-103} Our discussion of Thiele's reaction rate theory follows Ref. 54.

A. Phase space dividing surfaces: Definition and properties

Interpreting the bistable thermalizing mode^{74,75} in the model Hamiltonians of Sec. III as an isomerization coordinate, we see that the thermostat dynamics associated with Hamiltonian (3) is equivalent to an isomerization reaction at constant energy $\mathcal{H}=0$. We therefore consider the rate of unimolecular isomerization, at fixed energy, for a system described by a time-independent, n DOF classical Hamiltonian.

Points in the $2n$ -dimensional system phase space $\mathcal{M} = \mathbb{R}^{2n}$ are denoted $z \equiv (\pi, q) \in \mathcal{M}$. The system Hamiltonian is $\mathcal{H}(z)$, and the $(2n-1)$ dimensional energy shell at energy E , $\mathcal{H}(z)=E$, is denoted $\Sigma_E \subset \mathcal{M}$. The corresponding microcanonical phase space density is $\delta(E - \mathcal{H}(z))$, and the associated density of states for the complete energy shell at energy

E is

$$\rho(E) = \int_{\mathcal{M}} dz \delta(E - \mathcal{H}(z)). \quad (28)$$

In Appendix A, we provide analytical results for $\rho(E)$ for n -dimensional systems with Hamiltonians of the form (3), with $\Phi(q)$ an isotropic harmonic potential and n even.

The first step in the analysis is to define regions of the energy surface corresponding to reactant and product. For the isokinetic thermostat Hamiltonians considered here, there is a natural division of phase space into reactant region (say, $y < 0$) and product region $y > 0$. The DS between reactant and product is determined by symmetry to be the codimension-1 surface $y=0$, and we shall be concerned with the evaluation of the microcanonical reactive flux across this surface. (For fundamental work on reactive flux correlation functions and associated relaxation kinetics, see Refs. 104–108.)

For multidimensional systems such as polyatomic molecules ($n \geq 3$ DOF), it is *in general* not possible to define or compute a DS with desirable dynamical attributes such as the no-recrossing property by working in configuration space alone, and a phase space perspective is necessary.^{61–63,66,88,89,98–103}

As discussed in Sec. IV, Hamilton's equations for the thermostat Hamiltonian \mathcal{H} have an equilibrium of saddle-center-...-center stability type. The significance of saddle points of this type for Hamilton's equations is that, for a range of energies above that of the saddle, the energy surfaces have the *bottleneck property* in a phase space neighborhood near the saddle, i.e., the $2n-1$ dimensional energy surface *locally* has the geometrical structure of the product of a $2n-2$ dimensional sphere and an interval, $S^{2n-2} \times I$. In the vicinity of the bottleneck, we are able to construct a dividing surface depending on E , $DS(E)$, with very desirable properties: For each energy in this range above the saddle, $DS(E)$ locally “disconnects” the energy surface into two disjoint pieces with the consequence that the only way to pass from one piece of the energy surface to the other is to cross $DS(E)$. The DS has the geometrical structure of a $2n-2$ dimensional sphere, S^{2n-2} , which is divided into two $2n-2$ dimensional hemispheres, denoted $DS_{in}(E)$ and $DS_{out}(E)$ that are joined at an equator, which is a $2n-3$ dimensional sphere, S^{2n-3} . The hemisphere $DS_{in}(E)$ corresponds to initial conditions of trajectories that enter the reaction region while $DS_{out}(E)$ corresponds to initial conditions of trajectories that exit the reaction region, both by passing through the bottleneck in the energy surface. The equator S^{2n-3} is an invariant manifold of saddle stability type, a so-called NHIM.⁵⁷ The NHIM is of great physical significance: It is the actual “saddle” in phase space identified as the “activated complex” of reaction rate dynamics.^{66,109,110}

In the context of microcanonical rates, it has been shown that $DS_{in}(E)$ and $DS_{out}(E)$ have the essential *no-recrossing property* and that the flux across them is minimal.¹⁰⁰ We denote the directional flux across these hemispheres by $\phi_{in}(E)$ and $\phi_{out}(E)$, respectively, and note that $\phi_{in}(E) + \phi_{out}(E) = 0$. The magnitude of the flux is $|\phi_{in}(E)|$

$= |\phi_{out}(E)| \equiv \phi(E)$. Most significantly, the hemisphere $DS_{in/out}(E)$ is the correct surface across which to compute the “exact” flux into/out of the reaction region.

B. Phase space volumes and gap times

The disjoint regions of phase space corresponding to species A (reactant) and B (product) will be denoted \mathcal{M}_A and \mathcal{M}_B , respectively.¹¹¹

As discussed above, the DS can be rigorously defined to be locally a surface of no return (transition state). The microcanonical density of states for reactant species A is

$$\rho_A(E) = \int_{\mathcal{M}_A} dz \delta(E - H(z)), \quad (29)$$

with a corresponding expression for the density of states $\rho_B(E)$ for product B for the case of compact product energy shell.

For isokinetic thermostat Hamiltonians \mathcal{H} , the $\mathcal{H}=0$ energy surface extends to $\pm\infty$ in configuration space. Although the phase space volume $N(E)$ is finite for $E \rightarrow 0$, the corresponding density of states $\rho(E) = dN/dE$ may diverge as $E \rightarrow 0$. In Appendix A we derive analytical expressions for $\rho(E)$ associated with isotropic harmonic potentials $\Phi(\mathbf{q})$, $\mathbf{q} \in \mathbb{R}^n$, with n even. While $\rho(E)$ diverges as $E \rightarrow 0$ for $n=2$, it is finite in this limit for $n \geq 4$. Analytical expressions for $\rho(E)$ are not available for the model potentials studied here.

Provided that the flow is everywhere transverse to $DS_{in, out}(E)$, those phase points in the reactant region \mathcal{M}_A that lie on crossing trajectories^{51,112} (i.e., that will eventually react) can be specified uniquely by coordinates $(\bar{\mathbf{q}}, \bar{\boldsymbol{\pi}}, \psi)$, where $(\bar{\mathbf{q}}, \bar{\boldsymbol{\pi}}) \in DS_{in}(E)$ is a point on $DS_{in}(E)$, the incoming half of the DS, specified by $2(n-1)$ coordinates $(\bar{\mathbf{q}}, \bar{\boldsymbol{\pi}})$, and ψ is a time variable. (DSs constructed by normal form algorithms are guaranteed to be transverse to the vector field, except at the NHIM, where the vector field is tangent.^{61,98}) The point $\mathbf{z}(\bar{\mathbf{q}}, \bar{\boldsymbol{\pi}}, \psi)$ is reached by propagating the initial condition $(\bar{\mathbf{q}}, \bar{\boldsymbol{\pi}}) \in DS_{in}(E)$ forward for time ψ .^{54,82} As all initial conditions on $DS_{in}(E)$ (apart from a set of trajectories of measure zero lying on stable manifolds) will leave the reactant region in finite time by crossing $DS_{out}(E)$, for each $(\bar{\mathbf{q}}, \bar{\boldsymbol{\pi}}) \in DS_{in}(E)$ we can define the *gap time* $s = s(\bar{\mathbf{q}}, \bar{\boldsymbol{\pi}})$, which is the time it takes for the incoming trajectory to traverse the reactant region. That is, $\mathbf{z}(\bar{\mathbf{q}}, \bar{\boldsymbol{\pi}}, \psi = s(\bar{\mathbf{q}}, \bar{\boldsymbol{\pi}})) \in DS_{out}(E)$. For the phase point $\mathbf{z}(\bar{\mathbf{q}}, \bar{\boldsymbol{\pi}}, \psi)$, we therefore have $0 \leq \psi \leq s(\bar{\mathbf{q}}, \bar{\boldsymbol{\pi}})$.

The coordinate transformation $\mathbf{z} \rightarrow (E, \psi, \bar{\mathbf{q}}, \bar{\boldsymbol{\pi}})$ is canonical,^{82,86,87,113} so that the phase space volume element is

$$d^{2n}\mathbf{z} = dE d\psi d\sigma, \quad (30)$$

with $d\sigma \equiv d^{n-1}\bar{\mathbf{q}} d^{n-1}\bar{\boldsymbol{\pi}}$ an element of $2n-2$ dimensional area on the DS.

The magnitude $\phi(E)$ of the flux through dividing surface $DS(E)$ at energy E is given by

$$\phi(E) = \left| \int_{\text{DS}_{\text{in}}(E)} d\sigma \right|, \quad (31)$$

where the element of area $d\sigma$ is precisely the restriction to the DS of the appropriate flux $(2n-2)$ -form $\omega^{(n-1)}/(n-1)!$ corresponding to the Hamiltonian vector field associated with $H(z)$.^{100,114–116} The reactant phase space volume occupied by points initiated on the dividing surface DS_{in} with energies between E and $E+dE$ is therefore^{82,84–89}

$$dE \int_{\text{DS}_{\text{in}}(E)} d\sigma \int_0^s d\psi = dE \int_{\text{DS}_{\text{in}}(E)} d\sigma s, \quad (32a)$$

$$= dE \phi(E) \bar{s}, \quad (32b)$$

where the *mean gap time* \bar{s} is defined as

$$\bar{s} = \frac{1}{\phi(E)} \int_{\text{DS}_{\text{in}}(E)} d\sigma s, \quad (33)$$

and is a function of energy E . The reactant density of states $\rho_A^{\text{C}}(E)$ associated with crossing trajectories only (those trajectories that enter and exit the reactant region¹¹²) is then

$$\rho_A^{\text{C}}(E) = \phi(E) \bar{s}, \quad (34)$$

where the superscript C indicates the restriction to crossing trajectories. The result (34) is essentially the content of the so-called classical spectral theorem.^{84–89}

If *all* points in the reactant region of phase space eventually react (that is, all points lie on crossing trajectories^{51,112}) then

$$\rho_A^{\text{C}}(E) = \rho_A(E), \quad (35)$$

so that the crossing density of states is equal to the full reactant phase space density of states. Apart from a set of measure zero, all phase points $z \in \mathcal{M}_A$ can be classified as either trapped (T) or crossing (C).¹¹² A phase point in the trapped region \mathcal{M}_A^{T} never crosses the DS, so that the associated trajectory does not contribute to the reactive flux. Phase points in the crossing region \mathcal{M}_A^{C} do however eventually cross the DS, and so lie on trajectories that contribute to the reactive flux. In general, however, as a consequence of the existence of trapped trajectories (either trajectories on invariant *trapped* n -tori^{51,112} or trajectories asymptotic to other invariant objects of zero measure), we have the inequality^{82,112,117}

$$\rho_A^{\text{C}}(E) \leq \rho_A(E). \quad (36)$$

From the perspective of thermostat dynamics, the equality (35) is a *necessary condition* for ergodicity. If equality (35) does not hold, then there is a region of phase space of nonzero measure that is trapped on the reactant side of the DS.

If $\rho_A^{\text{C}}(E) < \rho_A(E)$, then it is in principle necessary to introduce corrections to statistical estimates of reaction rates.^{108,112,117–120} Numerical results for $\rho^{\text{C}}(E)$ and $\rho(E)$ for the HCN molecule are discussed in Refs. 54, 88, and 101.

C. Gap time and reactant lifetime distributions

The *gap time distribution* $\mathcal{P}(s; E)$ is of central interest in unimolecular kinetics:^{81,82} The probability that a phase point on $\text{DS}_{\text{in}}(E)$ at energy E has a gap time between s and $s+ds$ is equal to $\mathcal{P}(s; E)ds$. An important idealized gap distribution is the random, exponential distribution

$$\mathcal{P}(s; E) = k(E)e^{-k(E)s} \quad (37)$$

characterized by a single decay constant k (where k depends on energy E), with corresponding mean gap time $\bar{s} = k^{-1}$. An exponential distribution of gap times is taken to be a necessary condition for “statistical” behavior in unimolecular reactions.^{81–83,121}

The lifetime [time to cross the dividing surface $\text{DS}_{\text{out}}(E)$] of phase point $z(\bar{q}, \bar{p}, \psi)$ is $t = s(\bar{q}, \bar{p}) - \psi$, and the corresponding (normalized) reactant lifetime distribution function $\mathbb{P}(t; E)$ at energy E is^{81–83,121–124}

$$\mathbb{P}(t; E) = -\frac{d}{dt} \text{Prob}(t \geq t'; E)|_{t'=t}, \quad (38a)$$

$$= \frac{1}{\bar{s}} \int_t^{+\infty} ds \mathcal{P}(s; E), \quad (38b)$$

where the fraction of interesting (reactive) phase points having lifetimes between t and $t+dt$ is $\mathbb{P}(t; E)dt$.

Equation (38a) gives the general relation between the lifetime distribution and the fraction of trajectories having lifetimes greater than a certain value for arbitrary ensembles.^{122–124} Note that an exponential gap distribution (37) implies that the reactant lifetime distribution $\mathbb{P}(t; E)$ is also exponential;^{81,82,121–124} both gap and lifetime distributions for realistic molecular potentials have been of great interest since the earliest days of trajectory simulations of unimolecular decay, and many examples of nonexponential lifetime distributions have been found.^{68,119,122–128}

In the following section we discuss our numerical calculations of the following quantities for the isokinetic thermostat Hamiltonian at constant energy $\mathcal{H}=0$: Gap time distributions, mean gap time \bar{s} , reactive flux $\phi(E=0)$, reactive volume $\bar{s}\phi$, and reactant density of states, $\rho(E=0)$.

VI. NUMERICAL COMPUTATIONS FOR ISOKINETIC THERMOSTAT

A. Computations

In this section we discuss the computation of various dynamical quantities for the isokinetic thermostat Hamiltonians defined in Sec. III. The potentials for these isokinetic thermostat Hamiltonians with three and four DOF have the form of an exponentiated double well plus harmonic modes potential.

As already noted, for the cases examined here, we can exploit the symmetry of the potential functions. Thus, the DS between “reactant” and “product” is simply defined to be the symmetry plane $y=0$ for the double well potential. For more general nonsymmetric potentials, the DS in phase space can be computed using a normal form expansion.⁶⁶

1. System parameters

In addition to the frequencies characterizing the modes transverse to the reaction coordinate, it is necessary to choose values for the parameters $\beta \equiv 1/k_B T$, α , and ν . We adopt the following notation for presentation of our results: The system denoted by $H\beta\alpha\nu$ has three DOF and parameter values indicated, while the system $J\beta\alpha\nu$ has four DOF. We present numerical results for the three DOF systems H121, H321, H521 [Hamiltonian Eq. (20)] and four DOF systems J121, J321, J521 [Hamiltonian Eq. (23)].

In the units used here the height of the barrier to isomerization in the physical potential corresponds to $\beta=2$. The temperature values studied here therefore span a range of energy scales from well below the barrier height ($\beta=5$) to well above ($\beta=1$).

2. Computations

After choosing a set of parameters, we compute the following quantities (further details of computational methodology are given in Appendix B).

a. Gap time distribution and mean gap time. The distribution of gap times is obtained by starting trajectories on the forward DS and propagating them until they cross the backward DS. Initial conditions are obtained by uniform random sampling of the DS (see Appendix B). A discretized approximation to the gap time distribution $\mathcal{P}(s)$ is obtained by binning the gap times for the trajectory ensemble. The mean gap time \bar{s} is calculated as an unweighted average of computed gap times for the trajectory ensemble.

b. Lifetime distribution. The lifetime distribution $P(t)$ is obtained from the (discretized) gap time distribution by numerical integration [cf. Eq. (38)]. The form of the lifetime distribution is of interest: In particular, deviations from exponentiality are suggestive of nonstatistical dynamics.

The average lifetime $\langle t \rangle$ for the normalized lifetime distribution $P(\tau)$ is defined as

$$\langle t \rangle = \int_0^\infty dt P(t)t. \quad (39)$$

The random (exponential) lifetime distribution $P = \bar{k}e^{-\bar{k}t}$ extremizes the information entropy

$$S_P \equiv - \int_0^\infty dt P(t) \log[P(t)], \quad (40)$$

where $\bar{k} = \langle t \rangle^{-1}$ (see, for example, Ref. 129). One measure of the extent to which a calculated lifetime distribution $P(t)$ characterized by mean lifetime $\langle t \rangle$ deviates from exponentiality is the entropy deficit¹²⁹

$$\Delta S_P \equiv S_{\text{random}} - S(P), \quad (41a)$$

$$= 1 + \log \langle t \rangle + \int_0^\infty dt P(t) \log[P(t)]. \quad (41b)$$

c. Flux through DS. The flux through the DS is obtained by integrating the flux form $d\sigma$ [cf. Eq. (31)] over the DS. As the flux form is simply the phase space volume element as-

sociated with the “activated complex,” the reactive flux ϕ can be computed by uniform (random) sampling of the DS phase space (see Appendix B).

The associated reactive volume $2\phi \times \bar{s}$ is the total phase space volume on the energy shell $\mathcal{H}=0$ traced out by all the trajectories passing through the DS (in both directions, hence the factor of 2). Except for a set of measure zero, each trajectory returns to the DS, and only contributes to the reactive volume up until the gap time.

d. Reactant density of states. The classical density of states $\rho(E)$ of the reactant region at energy $\mathcal{H}=0$ is obtained by calculating a discretized approximation to $\rho(E)$ as a function of energy for $E < 0$, fitting $\rho(E)$ to a polynomial function in E , and evaluating the fitted $\rho(E)$ at $E=0$. The discretized approximation to $\rho(E)$ is obtained by randomly sampling phase points inside a suitably chosen hypercube, and binning the energies for sampled points with $E \leq 0$. (See Appendix B.)

Although in all cases the phase space volume (integrated density of states) $N(E)$ is found to be finite as $E \rightarrow 0$, for the three DOF isokinetic Hamiltonian [Eq. (20)] it is not clear from our numerical results whether or not the density of states actually diverges as $E \rightarrow 0$, and we have not been able to evaluate this limit analytically. For the isokinetic thermostat with $n=4$ DOF [Eq. (23)], our numerical results suggest that $\rho(E)$ is finite as $E \rightarrow 0$.

In Appendix A, we show analytically for isotropic harmonic potentials that, while $\rho(E)$ diverges logarithmically as $E \rightarrow 0$ for $n=2$ DOF, for $n \geq 4$ DOF, with n even, $\rho(E=0)$ is finite.

The value of the density of states $\rho(E)$ at $E=0$ is of some interest, as equality between the reactive density of states determined as the product of the mean gap time with the reactive flux and the total reactant density of states $\rho(E=0)$ is a necessary condition for ergodicity of the thermostat dynamics.

e. Thermostat dynamics. It is of course important to examine the effectiveness of the Hamiltonian isokinetic thermostats defined here as thermostats; that is, to assess how well time-averaged coordinate distributions evaluated over a single thermostat trajectory reproduce the Boltzmann distributions associated with the relevant temperature parameter β .

We therefore pick an initial condition at the coordinate origin on the DS with random momenta and $\mathcal{H}=0$, and propagate it for a long time ($t=20\,000$, many periods of the harmonic oscillator modes). Trajectories are integrated in MATHEMATICA (Ref. 130) using the function NDSolve with the SymplecticPartitionedRungeKutta method and fixed step size, $\Delta t=0.01$. By binning coordinate values over such a long trajectory, we obtain a discretized probability distribution function that can be compared with the desired canonical (Boltzmann) coordinate distribution.

Moments of powers x^k and y^k obtained as time averages over the trajectory can also be compared with thermal averages at temperature T .

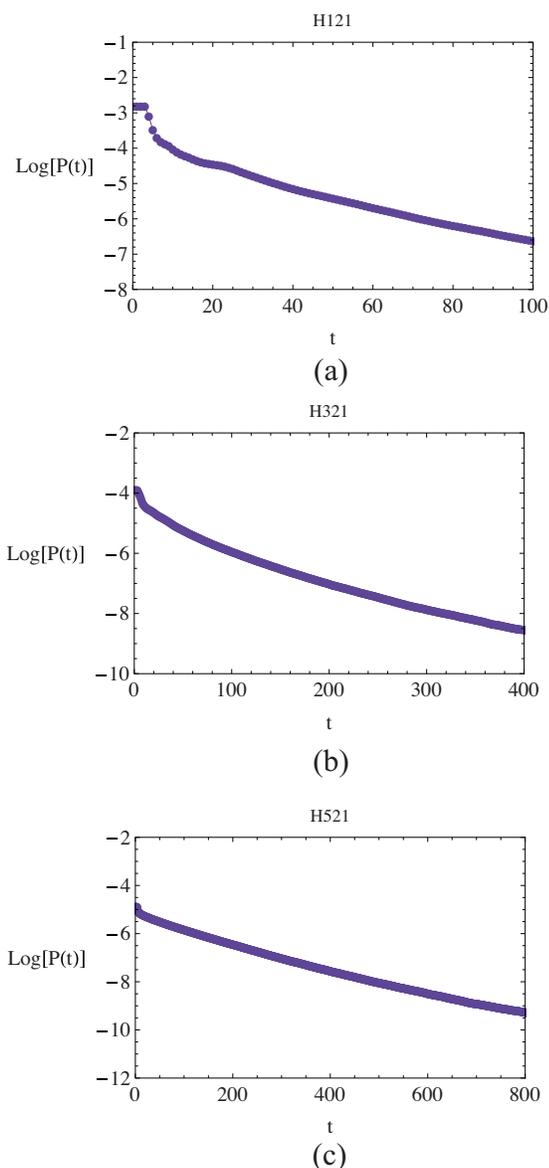


FIG. 2. Lifetime distributions for the three DOF Hamiltonian isokinetic thermostat. The lifetime distribution is derived from the distribution of gap times obtained by initiating trajectories on the incoming DS and propagating them until they cross the outgoing DS. (a) H121, $\beta=1$. (b) H321, $\beta=3$. (c) H521, $\beta=5$.

B. Numerical results: Three DOF

1. Lifetime distributions

Computed lifetime distributions for the three DOF thermostats H121, H321, and H521 are shown in Fig. 2. It can be

seen qualitatively from these plots of $\log[P(t)]$ versus t that, following initial transient decay, the lifetime distributions become more exponential as the temperature decreases (i.e., as β increases). The entropic measure of the deviation defined in Eq. (41) was computed for (renormalized) lifetime decay curves with transients excluded (that is, we take $t \geq \langle t \rangle$). The resulting values of ΔS are shown in Table I; the ΔS values reflect the qualitative observation that the lifetime distributions become more exponential as β increases.

2. Phase space volumes

Numerical values for the flux, mean gap time, and reactant phase space volumes are given in Table I. The magnitude of the reactive flux ϕ decreases with temperature (as β increases), as does the inverse gap time. Smaller inverse gap times therefore correlate with lifetime distributions that are more nearly exponential (cf. Ref. 51).

Note that energy surface volumes for the three DOF systems are not shown in Table I. The reason for this omission is that, on the basis of our numerical computations (results not shown here), we cannot be sure whether or not $\rho(E)$ diverges for three DOF systems as $E \rightarrow 0$ (cf. Appendix A). As we do not have an analytical proof that the density of states at $E=0$ is finite, we cannot rule out the possibility of a divergence.

3. Thermostat coordinate distributions

Coordinate distributions for the three DOF Hamiltonian isokinetic thermostat are shown in Figs. 3–5. The histogrammed distributions are obtained by time-averaging over a single trajectory, while the solid red curves are the associated canonical (Boltzmann) distributions. Moments $\langle q^k \rangle$ for $q = x_1$ are shown in Fig. 6.

Qualitatively, it is apparent from these results that the effectiveness of the Hamiltonian isokinetic thermostat, as judged by the similarity of trajectory-based and Boltzmann coordinate distributions and moments, increases with temperature. Recall that the lifetime distributions become more nearly exponential as temperature decreases.

Despite the possibility of an infinite energy shell volume, the Hamiltonian isokinetic thermostat for three DOF does in fact serve to thermalize the two uncoupled harmonic modes quite effectively. Note, however, that even for the long trajectories considered, numerical distributions obtained for the thermalizing (y) coordinate are not symmetric about $y=0$.

TABLE I. Computed mean gap times, fluxes, reactive phase space volumes, energy surface volumes, and entropy deficits for lifetime distributions for three DOF and four DOF model Hamiltonians. Details of the computations are discussed in Appendix B.

	Mean gap time	Flux	Reactive volume	Energy surface volume	ΔS
H121	16.57	6.978	231.28	...	0.034
H321	48.88	0.773	75.61	...	0.026
H521	130.41	0.280	72.99	...	0.019
J121	12.69	41.490	1053.36	1053.48	0.037
J321	38.51	1.536	118.31	118.66	0.021
J521	101.60	0.334	67.87	69.47	0.010

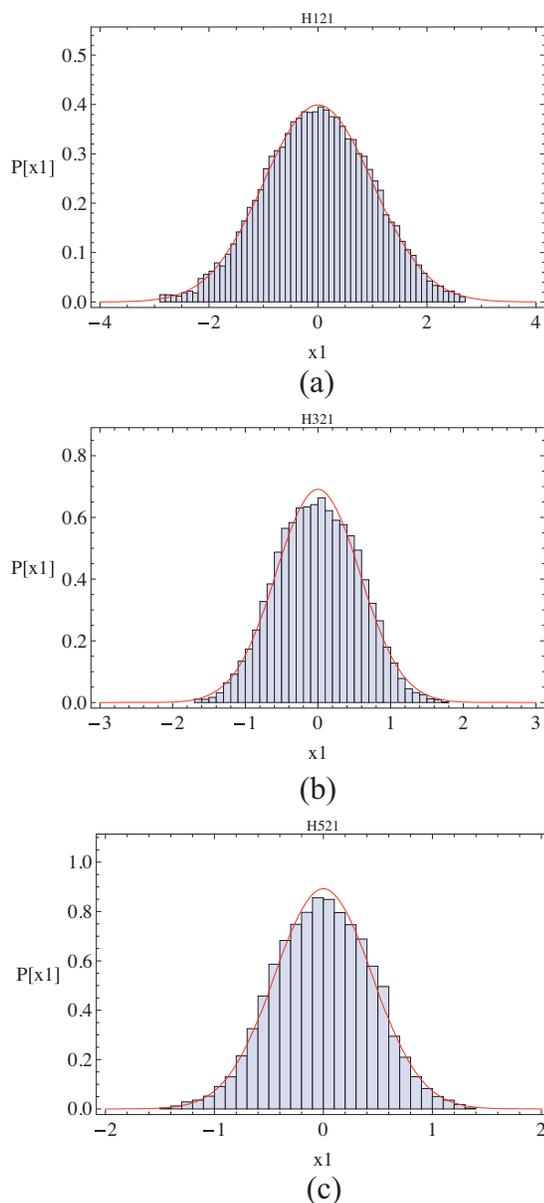


FIG. 3. Coordinate distributions for three DOF Hamiltonian isokinetic thermostat obtained by averaging over a single trajectory. Numerical distributions for the x_1 coordinate (histograms) are compared with the Boltzmann distribution (solid line). (a) H121, $\beta=1$. (b) H321, $\beta=3$. (c) H521, $\beta=5$.

C. Numerical results: Four DOF

1. Lifetime distributions

Computed lifetime distributions for the four DOF thermostats J121, J321, and J521 are shown in Fig. 7. Values of the lifetime distribution entropy deficit ΔS are shown in Table I. As for the three DOF systems, the four DOF lifetime distributions become more exponential as the temperature decreases.

2. Phase space volumes

Numerical values for the flux, mean gap time, and reactant phase space volumes for four DOF systems are given in Table I. Both the magnitude of the reactive flux ϕ and the

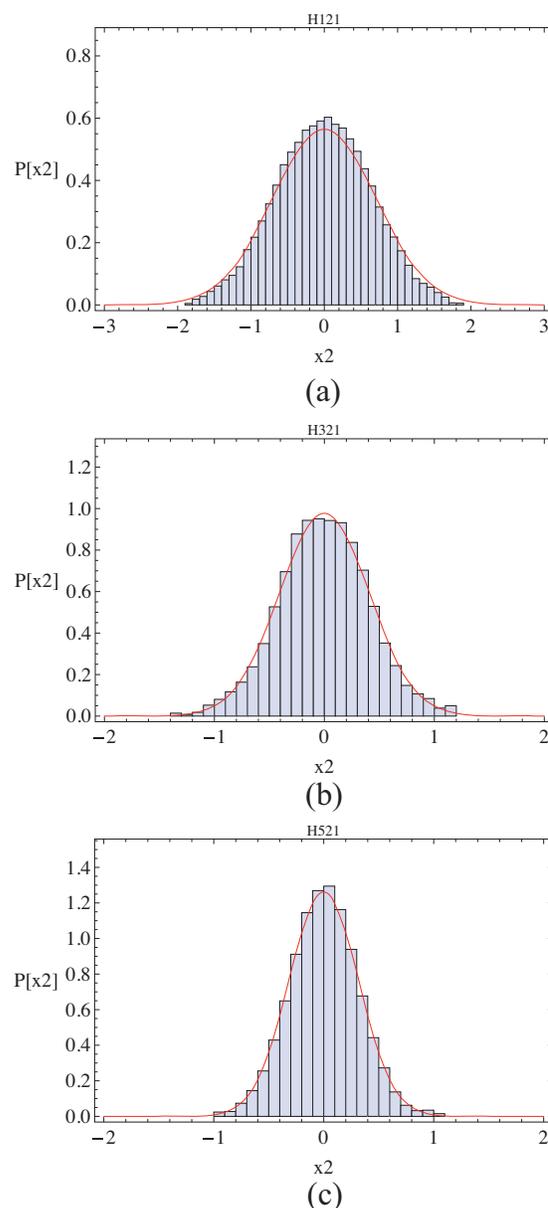


FIG. 4. Coordinate distributions for three DOF Hamiltonian isokinetic thermostat obtained by averaging over a single trajectory. Numerical distributions for the x_2 coordinate (histograms) are compared with the Boltzmann distribution (solid line). (a) H121, $\beta=1$. (b) H321, $\beta=3$. (c) H521, $\beta=5$.

inverse gap time decrease with temperature (as β increases), while the lifetime decay curves become more nearly exponential as β increases.

Comparison between reactive volumes and total energy surface volumes shows that both J121 and J321 systems satisfy (at least, within numerical error) the necessary condition for ergodicity, while the J521 system shows a minor deviation from equality.

3. Thermostat coordinate distributions

Distributions for coordinates x_1 and y computed for the four DOF Hamiltonian isokinetic thermostat are shown in Figs. 8 and 9, respectively. The histogrammed distributions are obtained by time-averaging over a single trajectory, while the solid red curves are the associated canonical (Boltzmann) distributions. Moments $\langle q^k \rangle$ for $q=x_1$ are shown in Fig. 10.

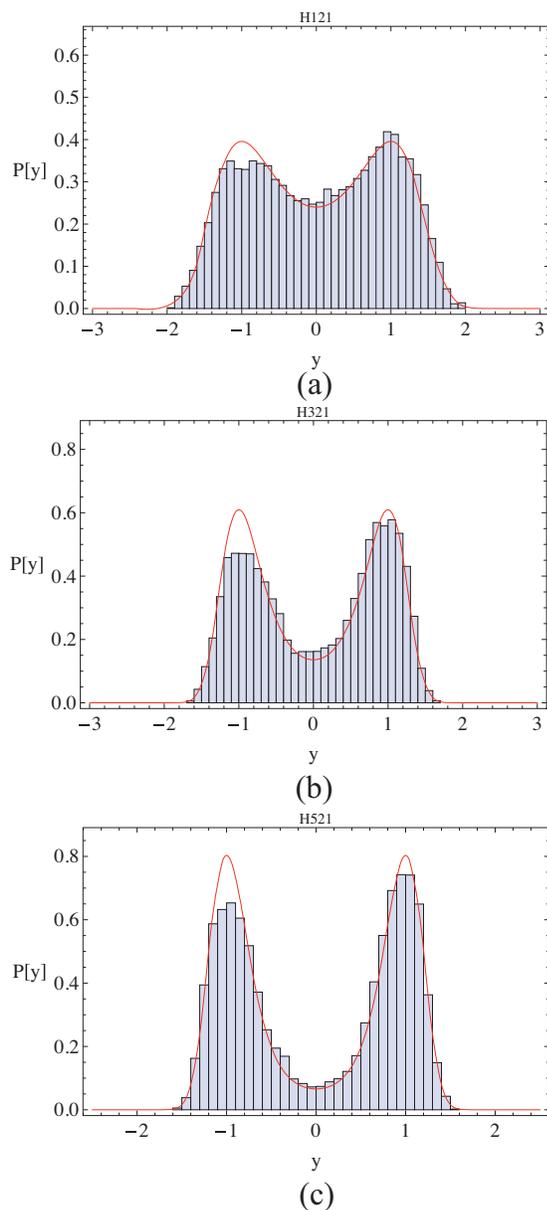


FIG. 5. Coordinate distributions for three DOF Hamiltonian isokinetic thermostat obtained by averaging over a single trajectory. Numerical distributions for the y coordinate (histograms) are compared with the Boltzmann distribution (solid line). (a) H121, $\beta=1$. (b) H321, $\beta=3$. (c) H521, $\beta=5$.

As for three DOF, the effectiveness of the Hamiltonian isokinetic thermostat, as judged by the similarity of trajectory-based and Boltzmann coordinate distributions and moments for the harmonic oscillator coordinates, increases with temperature. Recall that the lifetime distributions become more nearly exponential as temperature decreases.

VII. SUMMARY AND CONCLUSIONS

In this paper we have investigated the phase space structure and dynamics of a Hamiltonian isokinetic thermostat. By design, ergodic thermostat trajectories at fixed (zero) energy generate a canonical distribution in configuration space.^{2,54,78,79} The physical potentials studied consist of

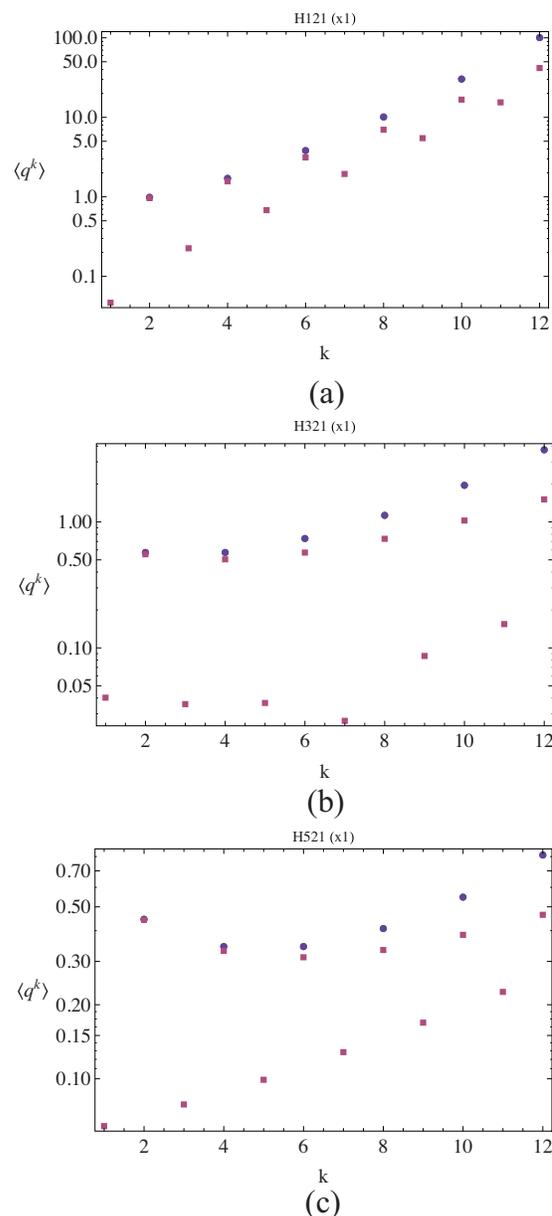


FIG. 6. Moments of the distribution of the coordinate x_1 obtained using the three DOF Hamiltonian isokinetic thermostat (squares) are compared with those for the Boltzmann distribution (circles). Odd moments are identically zero. (a) H121, $\beta=1$. (b) H321, $\beta=3$. (c) H521, $\beta=5$.

a single bistable mode (the thermalizing DOF^{74,75}) plus transverse harmonic modes. Although these modes are not coupled in the physical potential, the potential for the Hamiltonian thermostat is obtained by exponentiation of the physical potential, which introduces coupling between the modes.

Interpreting the bistable mode as a reaction (isomerization) coordinate, we are able to establish connections with the theory of unimolecular reaction rates,^{45,48,121,126} in particular the formulation of isomerization rates in terms of gap times.^{81,82} In Thiele's general formulation,⁸² the gap time is the time taken for a reactive trajectory initiated on a DS in phase space to return to the surface. (Such phase space DSs in multidimensional systems have been defined and computed using normal form theory.⁶¹⁻⁶⁶) In the molecular context, the distribution of gap times for a microcanonical en-

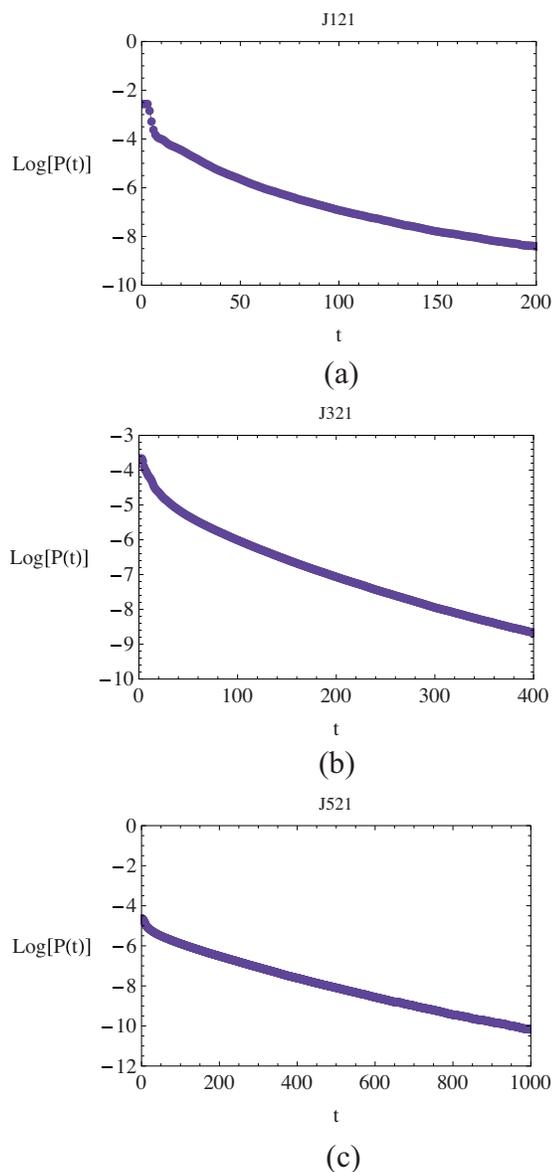


FIG. 7. Lifetime distributions for the four DOF Hamiltonian isokinetic thermostat. The lifetime distribution is derived from the distribution of gap times obtained by initiating trajectories on the incoming DS and propagating them until they cross the outgoing DS. (a) J121, $\beta=1$. (b) J321, $\beta=3$. (c) J521, $\beta=5$.

semble initiated on the DS is of great dynamical significance; an exponential distribution of the lifetimes for the reactive ensemble is usually taken to be an indicator of statistical behavior. Moreover, comparison of the magnitude of the phase space volume swept out by reactive trajectories as they pass through the reactant region with the total phase space volume (classical density of states) for the reactant region provides a necessary condition for ergodic dynamics. If the total density of states is appreciably larger than the reactive volume, the system cannot be ergodic. Ergodicity is a necessary condition for a thermostat to generate the correct phase space distribution; our use of a necessary condition for ergodicity derived via the spectral theorem establishes a significant connection between concepts of reaction rate theory and the properties of the Hamiltonian isokinetic thermostats studied here.

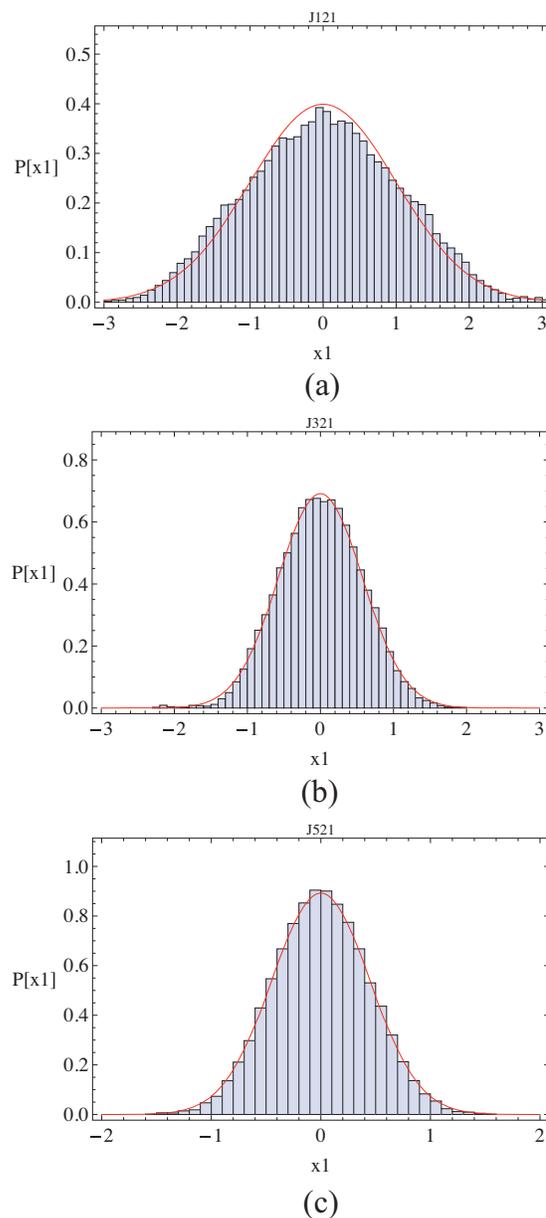


FIG. 8. Coordinate distributions for four DOF Hamiltonian isokinetic thermostat obtained by averaging over a single trajectory. Numerical distributions for the x_1 coordinate (histograms) are compared with the Boltzmann distribution (solid line). (a) J121, $\beta=1$. (b) J321, $\beta=3$. (c) J521, $\beta=5$.

We have computed gap times, associated lifetime distributions, mean gap times, reactive fluxes, reactive volumes, and total reactant phase space volumes for model systems with three and four DOF. The symmetry of the model potentials studied means that in all cases the DS is defined by a single condition on the thermalizing coordinate, $y=0$. The thermostats were studied at three different temperatures. For four DOF, the necessary condition for ergodicity is approximately satisfied at all three temperatures. For both three and four DOF systems, nonexponential lifetime distributions are found at high temperatures ($\beta=1$, where the potential barrier to isomerization is $1/2$ in the same units), while at low temperatures ($\beta=5$) the lifetime distribution is more nearly exponential.

We have quantified the degree of exponentiality of the lifetime distribution by computing the information entropy

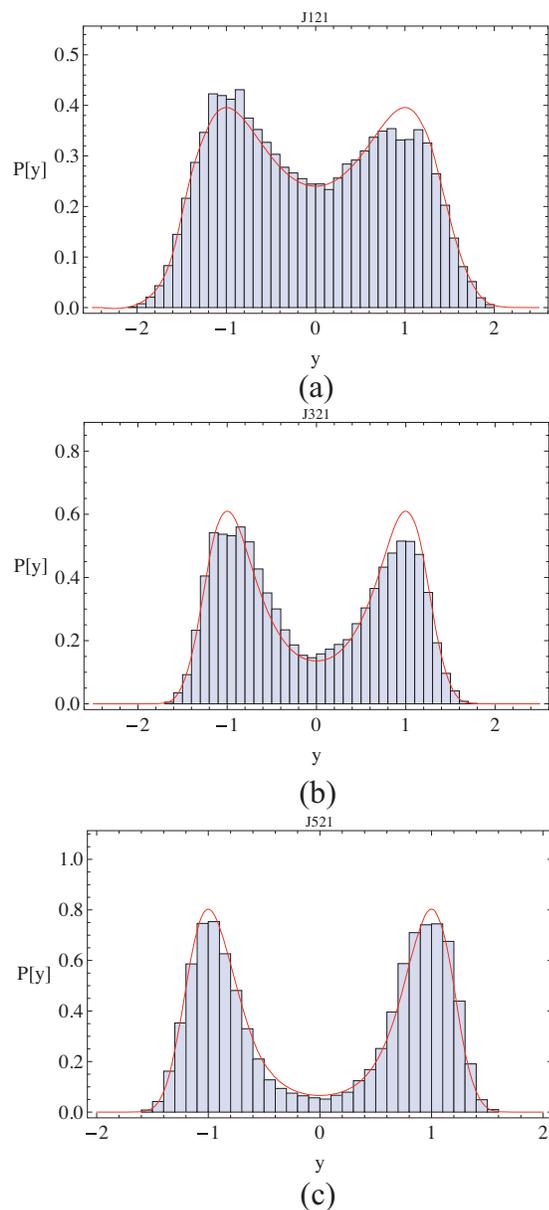


FIG. 9. Coordinate distributions for four DOF Hamiltonian isokinetic thermostat obtained by averaging over a single trajectory. Numerical distributions for the y coordinate (histograms) are compared with the Boltzmann distribution (solid line). (a) J121, $\beta=1$. (b) J321, $\beta=3$. (c) J521, $\beta=5$.

deficit with respect to pure exponential decay. From the standpoint of unimolecular reaction rate theory, the decay becomes more statistical at lower T (smaller flux). This finding is in accord with the early observations of De Leon and Berne⁵¹ on isomerization dynamics in a model two-mode system. However, kinetic parameters derived from the gap distributions we compute for our model thermostats have no direct connection with rate processes occurring in the physical system of interest; the analysis of lifetimes, etc., is simply a means to characterize thermostat dynamics.

We have examined the efficacy of the Hamiltonian isokinetic thermostat by computing coordinate distributions averaged over a single long trajectory initiated at random on the DS. For the parameter values used here, coordinate distributions are more nearly canonical (Boltzmann-like) at lower temperatures.

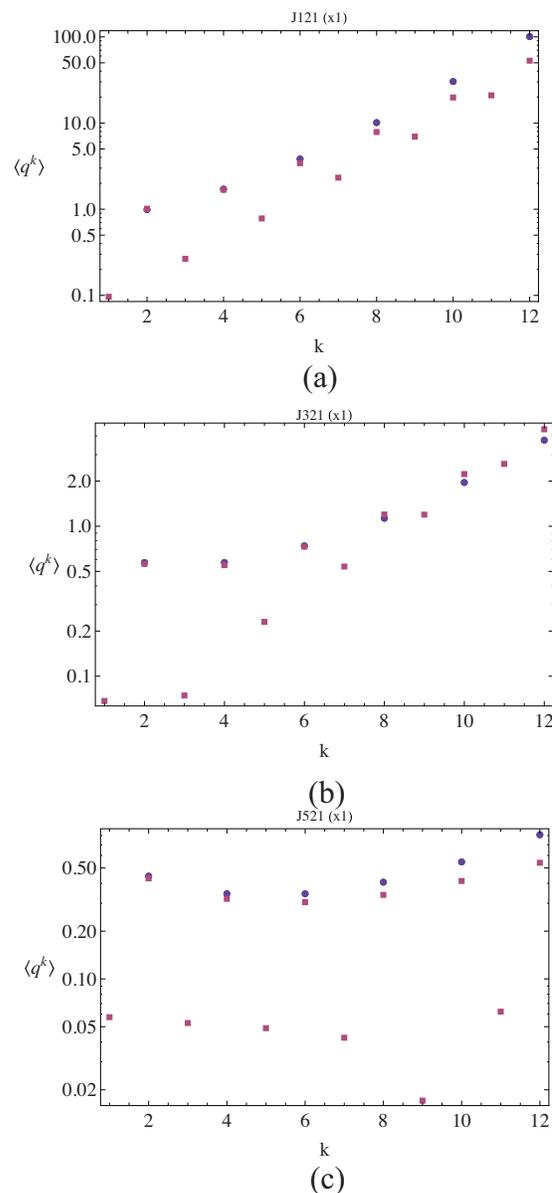


FIG. 10. Moments of the distribution of the coordinate x_1 obtained using the four DOF Hamiltonian isokinetic thermostat (squares) are compared with those for the Boltzmann distribution (circles). Odd moments for the Boltzmann distribution are identically zero. (a) J121, $\beta=1$. (b) J321, $\beta=3$. (c) J521, $\beta=5$.

It remains for future work to establish more quantitative correlations between dynamical attributes of the Hamiltonian thermostat and thermostat effectiveness.

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APPENDIX A: PHASE SPACE VOLUME AND CLASSICAL DENSITY OF STATES—ANALYTICAL RESULTS FOR HARMONIC POTENTIALS

An important question arising in numerical investigation of the dynamics and phase space structure of Hamiltonian

isokinetic thermostats concerns the behavior of the density of states $\rho(E)$ in the limit $E \rightarrow 0$. Specifically, it is important to know whether or not the density of states diverges in this limit. It can be difficult to establish convergence of $\rho(E)$ to a finite value as $E \rightarrow 0$ on the basis of numerical calculations of the phase space volume $N(E)$.

In this Appendix, we evaluate analytically the phase space volume $N(E)$ and the associated density of states $\rho(E) = dN(E)/dE$ for a Hamiltonian of the form

$$\mathcal{H} = \frac{1}{2} \bar{\pi}^2 - \frac{1}{2} e^{-\Phi}, \quad (\text{A1})$$

with

$$\Phi(q) = \frac{1}{2} \left[\sum_{j=1}^n q_j^2 \right], \quad (\text{A2})$$

and even dimensionality n . Physically, the potential (A2) corresponds to n uncoupled degenerate harmonic oscillators; we determine the phase space volume for the exponentiated harmonic potential appearing in the Hamiltonian (A1).

We start by evaluating the phase space volume $N(E)$ enclosed by the energy shell $\mathcal{H} = E \leq 0$. The potential (A2) is spherically symmetric, and so depends only on r , the radial coordinate in n -dimensional configuration space. Let $\bar{r} = \bar{r}(E)$ be the value of r for which the exponentiated harmonic potential is equal to E ,

$$\bar{r} = \sqrt{-2 \log[-2E]}. \quad (\text{A3})$$

Note that, as $E \rightarrow 0$ from below, $\bar{r}(E) \rightarrow \infty$. At fixed r , $0 \leq r \leq \bar{r}$, the magnitude of the momentum $|\bar{\pi}|$ is

$$|\bar{\pi}| = \sqrt{2E + e^{-r^2/2}}, \quad (\text{A4a})$$

$$= e^{-r^2/4} \sqrt{e^{(r^2 - \bar{r}^2)/2} - 1}. \quad (\text{A4b})$$

If $\mathcal{V}(x, n)$ is the volume of a ball of radius x in n dimensions,

$$\mathcal{V}(x, n) = \frac{\pi^{n/2} x^n}{\Gamma\left[\frac{n}{2} + 1\right]}, \quad (\text{A5})$$

then the total phase space volume $N(E)$ is given by the integral

$$N(E) = \int_0^{\bar{r}(E)} dr \mathcal{V}(|\bar{\pi}|, n) \frac{d\mathcal{V}(r, n)}{dr}, \quad (\text{A6a})$$

$$= \frac{\pi^n}{\Gamma\left[\frac{n}{2} + 1\right]^2} \mathcal{I}_n, \quad (\text{A6b})$$

with

$$\mathcal{I}_n \equiv \int_0^{\bar{r}(E)} dr n r^{n-1} e^{-nr^2/4} [e^{(r^2 - \bar{r}^2)/2} - 1]^{n/2}. \quad (\text{A7})$$

For even dimension n , this integral can be evaluated explicitly.

For $n=2$ DOF, the integral \mathcal{I}_2 is

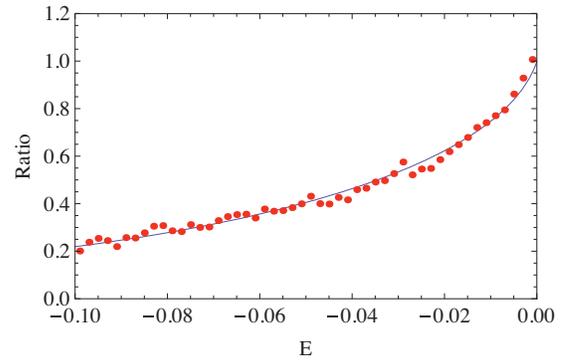


FIG. 11. Comparison of the numerically determined $\rho(E)$ for the four DOF harmonic potential with the theoretical expression (A14) over the energy range $-0.1 \leq E \leq 0$. The phase space volume $N(E)$ was determined by random sampling of a phase space hypercube using $N_p = 5 \times 10^7$ points. All $\rho(E)$ are divided by $8\pi^4$ [cf. Eq. (A15)].

$$\mathcal{I}_2 = 2 - e^{-\bar{r}^2/2} (2 + \bar{r}^2). \quad (\text{A8})$$

In terms of the energy E , we have

$$N(E) = 2\pi^2 (1 + E(2 - 2 \log[-2E])), \quad (\text{A9})$$

and the limiting value of the phase space volume is therefore finite

$$\lim_{E \rightarrow 0} N(E)|_{n=2} = 2\pi^2. \quad (\text{A10})$$

Despite the fact that, when $N(E)$ is plotted as a function of E , the derivative of $N(E)$ appears to be finite at $E=0$, the density of states $\rho(E)$ is

$$\rho(E) = \frac{dN(E)}{dE} = -4\pi^2 \log[-2E], \quad (\text{A11})$$

which diverges as $E \rightarrow 0$. This divergence is difficult to identify in a plot of $N(E)$ versus E even when the form of $N(E)$ is known, and is also difficult to see in plots of numerically determined $\rho(E)$.

For $n=4$,

$$\mathcal{I}_4 = e^{-\bar{r}^2} (\bar{r}^4 + 6\bar{r}^2 - 16e^{\bar{r}^2/2} + 14) + 2. \quad (\text{A12})$$

In terms of E , we have

$$N(E) = \frac{1}{2} \pi^4 (1 + 16E + 28E^2 - 24E^2 \log[-2E] + 8E^2 \log[-2E]^2), \quad (\text{A13})$$

and the associated $\rho(E)$ is

$$\rho(E) = 8\pi^4 (1 + 2E - 2E \log[-2E] + E \log[-2E]^2), \quad (\text{A14})$$

which is finite as $E \rightarrow 0$. In fact, we have

$$\lim_{E \rightarrow 0} \rho(E) = 8\pi^4 \approx 779.27. \quad (\text{A15})$$

Figure 11 compares the ratio of the numerically determined $\rho(E)$ for the four DOF harmonic potential to the theoretical expression (A14), over the energy range $-0.1 \leq E \leq 0$. The numerical $\rho(E)$ is obtained by randomly sampling phase space points inside a suitably chosen hypercube, and binning points according to the value of the Hamiltonian. The result-

ing histogram provides a discretized approximation to $\rho(E)$.

The value of $\rho(E=0)$ for the 4D harmonic oscillator potential obtained using the procedure discussed in Appendix B is 773.43, which is within 1% of the exact result (A15).

To see the structure of these expressions in the general case for $n=2m$, m integer, note that the integral

$$\mathcal{I}_{2m} \equiv \int_0^{\bar{r}(E)} dr 2mr^{2m-1} e^{-m\bar{r}^2/2} [e^{(\bar{r}^2-r^2)/2} - 1]^m \quad (\text{A16})$$

is a sum of terms of the form (c_j constant)

$$\mathcal{J}_j = c_j e^{-(m-j)\bar{r}^2/2} \int_0^{\bar{r}(E)} dr r^{2m-1} e^{-jr^2/2}, \quad j=0, 1, \dots, m. \quad (\text{A17})$$

The only term that contributes to $N(E)$ in the limit $E \rightarrow 0$ ($\bar{r} \rightarrow \infty$) is that with $j=m$, so that

$$\begin{aligned} \lim_{E \rightarrow 0} N(E) &= \frac{\pi^{2m} 2m}{\Gamma[m+1]^2} \int_0^\infty dr r^{2m-1} e^{-mr^2/2} \\ &= \frac{\pi^{2m}}{\Gamma[m+1]} \left(\frac{2}{m}\right)^m. \end{aligned} \quad (\text{A18})$$

The limiting value of the phase space volume is therefore finite for all positive even integers $n=2m$.

To find the general form for the density of states, $\rho(E)$, it is necessary to evaluate $d\mathcal{I}_{2m}/dE$. Differentiating the integral, noting that the integrand vanishes for $r=\bar{r}(E)$, and using

$$\frac{dE}{d\bar{r}} = \frac{\bar{r}}{2} e^{-\bar{r}^2/2}, \quad (\text{A19})$$

we have

$$\frac{d\mathcal{I}_{2m}}{dE} = \int_0^{\bar{r}(E)} dr (2m)^2 r^{2m-1} e^{-(m-1)\bar{r}^2/2} [e^{(\bar{r}^2-r^2)/2} - 1]^{m-1}. \quad (\text{A20})$$

It is clear that $\rho(E)$ will diverge as $E \rightarrow 0$ only for the case $m=1$; for all other positive integer m , the derivative $dN(E)/dE$ converges to a finite value as $E \rightarrow 0$.

We have not been able to derive corresponding analytical expressions for $\rho(E)$ for the isotropic harmonic case for odd dimensions.

APPENDIX B: PHASE SPACE SAMPLING AND COMPUTATION OF DYNAMICAL QUANTITIES FOR ISOKINETIC THERMOSTATS

In this Appendix we provide additional details of the computations discussed in Sec. VI. The potentials for the isokinetic thermostat Hamiltonians discussed in Sec. III are obtained by exponentiating the physical potential, which has the form of a one dimensional double well plus uncoupled harmonic modes. For comparison with the analytical results for the density of states for an exponentiated isotropic harmonic potential given in Appendix A, we have also investigated the corresponding Hamiltonians numerically for 2, 3, and 4 DOF.

As mentioned in Sec. VI, the symmetry of our Hamiltonians means that the DS between reactant and product is simply defined to be the symmetry plane $y=0$ for the double well potential. The results reported in this paper were obtained using algorithms coded in PYTHON;¹³¹ calculation efficiency was not a prime objective and could no doubt be improved significantly. Many of these calculations required large numbers of samples and were therefore subject to a trade-off between sample size and accuracy. The size of production runs was estimated on the basis of results obtained from shorter trial runs.

At the value of the Hamiltonian $\mathcal{H}=0$, the energy surface extends to infinity in coordinate (\mathbf{q}) space; the energy of the saddle point is negative. As the exponentiated potential is bounded from below, momentum components are bounded but coordinate components are unbounded. For phase space points with larger coordinate component magnitudes, the relative probability of having energy $E \leq 0$ decreases. A phase space sampling region (“box”) is therefore used with appropriate ranges of the momentum components and an upper limit on coordinate components chosen to be large enough that the (small) neglected volume associated with points having $E \leq 0$ does not affect the accuracy of the result to within a chosen tolerance at the sampling density used. The largest values for coordinate component magnitudes were taken to be either 4.0 or 5.0 (in the units used here).

Phase space points are randomly sampled within the box, and the required energy surface volume calculated by estimating the fraction of points which satisfy the appropriate energy condition. Increased accuracy requires both a large number of sample points and a large box; there is clearly a trade-off between calculation accuracy, box size (hence neglected volume outside the box), sample size, and processing time. We have aimed for an accuracy of 1% while maintaining as far as possible a common approach across all the parameter sets studied. A count was taken of points in the periphery (outer 10% of coordinates) of the box to check on the validity of this process.

Details of the computations now follow.

1. Mean gap time

The mean gap time is calculated as an appropriately weighted average over the DS. We sample the phase space manifold defined by constraints $y=0$, $p_y=0$. Other coordinate and momentum components (x, p_x) were randomly sampled within a chosen box. The value of the Hamiltonian was calculated at each point, and sample points with energies $E > 0$ were rejected. Points with $E \leq 0$ were used to seed a trajectory from the DS. The value of the saddle point momentum p_y was calculated (with $p_y \geq 0$) so that the total energy $\mathcal{H}=0$, giving a trajectory initial condition on the DS. Note that the momentum p_y is only calculated after a point on the DS has been accepted. The trajectory was then integrated using a fourth-order Runge–Kutta algorithm until it recrossed the surface $y=0$. The variation of the energy along the integrated trajectory was monitored to check the stability of the integration procedure. The maximum cutoff time for the trajectory was either 2000 or 5000 time units (for differ-

ent parameter sets). The gap time for each trajectory was recorded and the average calculated over $\sim 100\,000$ trajectories.

2. Flux

The directional flux ϕ is the volume of the DS $y=0$ at $\mathcal{H}=0$ [cf. Eq. (31)]. It is calculated by uniformly sampling the restricted phase space region as above and again calculating the value of the Hamiltonian to decide whether each sampled point is within the DS (that is, has energy ≤ 0). The fraction of points accepted times the volume of the sampling region yields the volume of the DS for $\mathcal{H}=0$.

3. Reactive volume

The reactive volume is obtained by calculating (twice) the product of the mean gap time \bar{s} and the one-way flux ϕ . Since only positive values of the saddle point momentum are used, only trajectories on the positive side of the DS are used to calculate the mean gap time. For symmetric systems such as those treated here, an additional factor of 2 is all that is required to calculate the total reactive volume on both sides of the DS.

4. Energy surface volume and density of states

We wish to calculate the density of states $\rho(E)$ at energy $E=0$, where $\rho(E)$ is the derivative with respect to energy of $N(E)$, the total phase space volume enclosed with the energy shell $\mathcal{H}=0$. In order to do so we determine the volume of a set of energy shells of thickness ΔE . These data can then be used to obtain a polynomial fit to $\rho(E)$ and hence its value at $E=0$.

A phase space sampling region (box) of full phase space dimensionality is used, so that we sample in two more dimensions than the computations described above. We sample phase space points $z \equiv (\pi, q) \in \mathcal{M}$ with $\mathcal{H}(z) \leq 0$, and consider the region of phase space occupied by points with energies $-k \times \Delta E \leq E \leq -(k-1) \times \Delta E$, with ΔE typically 10^{-5} and k a positive integer. The volume of each of these energy shells is estimated by determining the number of points inside it as a fraction of those inside the full box; the corresponding density of states $\rho(E_k)$ at $E_k = -(k - \frac{1}{2}) \times \Delta E$ can then be obtained by dividing the volume of the k th shell by ΔE .

The density of states $\rho(E)$ can then be calculated using either of two approaches. The total volume of a set of k_{\max} such shells can be calculated, and the resulting energy volume $N(E)$ for $-k_{\max} \times \Delta E \leq E \leq 0$ fitted to, for example, a polynomial in E . The derivative of the fitting function can then be evaluated at $\Delta E=0$. An alternative procedure is to fit directly the estimated values of $\rho(E)$ to a polynomial in E and use the resulting fit to obtain $\rho(E)$ at $E=0$. Both methods should in principle give the same result.

In the calculations reported here, $\rho(E)$ was calculated for a set of 2×10^3 shells with $\Delta E=10^{-5}$, and a fifth-order

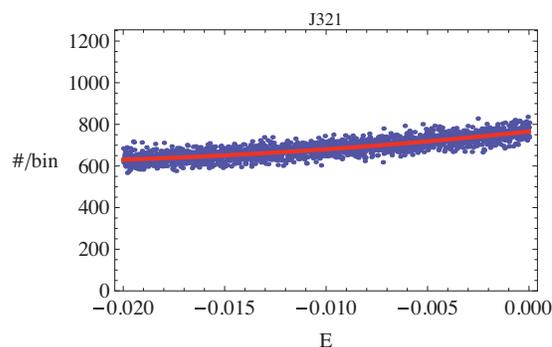


FIG. 12. Number of points per energy bin vs energy E (width $\Delta E=10^{-5}$, $-0.02 \leq E \leq 0$) for the J321 four DOF Hamiltonian. The red curve is a fit to the numerical data using a fifth-order polynomial. The fit to the data yields a value $\rho(E=0)=118.66$.

polynomial fit to $\rho(E)$ calculated over the range of $-0.02 \leq E \leq 0$. Noise in the fitted data (see Fig. 12) can be reduced by smoothing (concatenating several energy intervals ΔE), and we have verified that the values of $\rho(E=0)$ we obtain are robust with respect to such smoothing.

- ¹ S. Nosé, *Prog. Theor. Phys. Suppl.* **103**, 1 (1991).
- ² G. P. Morriss and C. P. Dettmann, *Chaos* **8**, 321 (1998).
- ³ W. G. Hoover, K. Aoki, C. G. Hoover, and S. V. D. Groot, *Physica D* **187**, 253 (2004).
- ⁴ B. Leimkuhler and S. Reich, *Simulating Hamiltonian Dynamics* (Cambridge University Press, Cambridge, 2004).
- ⁵ P. Hunenberger, *Adv. Polym. Sci.* **173**, 105 (2005).
- ⁶ S. D. Bond and B. J. Leimkuhler, *Acta Numerica* **16**, 1 (2007).
- ⁷ D. J. Evans and G. P. Morriss, *Statistical Mechanics of Nonequilibrium Liquids* (Academic, New York, 1990).
- ⁸ W. G. Hoover, *Computational Statistical Mechanics* (Elsevier, New York, 1991).
- ⁹ C. J. Mundy, S. Balasubramanian, K. Bagchi, M. E. Tuckerman, G. J. Martyna, and M. L. Klein, *Nonequilibrium Molecular Dynamics*, Reviews in Computational Chemistry Vol. 14 (Wiley, New York, 2000), pp. 291–397.
- ¹⁰ M. Romero-Bastida and J. F. Aguilar, *J. Phys. A* **39**, 11155 (2006).
- ¹¹ W. G. Hoover and C. G. Hoover, *J. Chem. Phys.* **126**, 164113 (2007).
- ¹² O. G. Jepps and L. Rondoni, *J. Phys. A* **43**, 133001 (2010).
- ¹³ S. Nosé, *J. Chem. Phys.* **81**, 511 (1984).
- ¹⁴ W. G. Hoover, *Phys. Rev. A* **31**, 1695 (1985).
- ¹⁵ C. P. Dettmann and G. P. Morriss, *Phys. Rev. E* **54**, 2495 (1996).
- ¹⁶ C. P. Dettmann and G. P. Morriss, *Phys. Rev. E* **55**, 3693 (1997).
- ¹⁷ C. P. Dettmann, *Phys. Rev. E* **60**, 7576 (1999).
- ¹⁸ S. D. Bond, B. J. Leimkuhler, and B. B. Laird, *J. Comput. Phys.* **151**, 114 (1999).
- ¹⁹ M. E. Tuckerman, C. J. Mundy, and G. J. Martyna, *Europhys. Lett.* **45**, 149 (1999).
- ²⁰ M. E. Tuckerman, Y. Liu, G. Ciccotti, and G. J. Martyna, *J. Chem. Phys.* **115**, 1678 (2001).
- ²¹ A. Sergi and M. Ferrario, *Phys. Rev. E* **64**, 056125 (2001).
- ²² A. Sergi, *Phys. Rev. E* **67**, 021101 (2003).
- ²³ G. S. Ezra, *J. Math. Chem.* **35**, 29 (2004).
- ²⁴ V. E. Tarasov, *J. Phys. A* **38**, 2145 (2005).
- ²⁵ A. Sergi and P. V. Giaquinta, *J. Stat. Mech.: Theory Exp.* **2007**, P02013.
- ²⁶ A. Sergi and G. S. Ezra, *Phys. Rev. E* **81**, 036705 (2010).
- ²⁷ M. P. Wojtkowski and C. Liverani, *Commun. Math. Phys.* **194**, 47 (1998).
- ²⁸ P. Choquard, *Chaos* **8**, 350 (1998).
- ²⁹ G. J. Martyna, M. L. Klein, and M. E. Tuckerman, *J. Chem. Phys.* **97**, 2635 (1992).
- ³⁰ D. Kuznezov, A. Bulgac, and W. Bauer, *Ann. Phys. (N.Y.)* **204**, 155 (1990).
- ³¹ S. D. Bond and B. J. Leimkuhler, *Numer. Algorithms* **19**, 55 (1998).
- ³² *Singularities in Gravitational Systems: Applications to Chaotic Transport in the Solar System*, edited by D. Benest and C. Froeschlé (Springer, New York, 2002).

- ³³ J. M. Sanz-Serna and M. P. Calvo, *Numerical Hamiltonian Problems* (Chapman and Hall, London, 1994).
- ³⁴ E. Hairer, C. Lubich, and G. Wanner, *Geometric Numerical Integration: Structure Preserving Algorithms for Ordinary Differential Equations* (Springer, New York, 2002).
- ³⁵ I. P. Cornfeld, S. V. Fomin, and Y. G. Sinai, *Ergodic Theory* (Springer-Verlag, New York, 1982).
- ³⁶ R. Sturman, J. M. Ottino, and S. Wiggins, *The Mathematical Foundations of Mixing* (Cambridge University Press, New York, 2006).
- ³⁷ V. L. Golo, V. N. Salnikov, and K. V. Shaitan, *Phys. Rev. E* **70**, 046130 (2004).
- ³⁸ H. Watanabe and H. Kobayashi, *Mol. Simul.* **33**, 77 (2007).
- ³⁹ H. Watanabe and H. Kobayashi, *Phys. Rev. E* **75**, 040102 (2007).
- ⁴⁰ F. Legoll, M. Luskin, and R. Moeckel, *Arch. Ration. Mech. Anal.* **184**, 449 (2007).
- ⁴¹ F. Legoll, M. Luskin, and R. Moeckel, *Nonlinearity* **22**, 1673 (2009).
- ⁴² B. J. Leimkuhler and C. R. Sweet, *SIAM J. Appl. Dyn. Syst.* **4**, 187 (2005).
- ⁴³ P. F. Tupper, *SIAM J. Appl. Dyn. Syst.* **4**, 563 (2005).
- ⁴⁴ R. D. Skeel, *SIAM J. Sci. Comput. (USA)* **31**, 1363 (2009).
- ⁴⁵ P. J. Robinson and K. A. Holbrook, *Unimolecular Reactions* (Wiley, New York, 1972).
- ⁴⁶ R. G. Gilbert and S. C. Smith, *Theory of Unimolecular and Recombination Reactions* (Blackwell Scientific, Oxford, 1990).
- ⁴⁷ T. Baer and W. L. Hase, *Unimolecular Reaction Dynamics* (Oxford University Press, New York, 1996).
- ⁴⁸ W. Forst, *Unimolecular Reactions* (Cambridge University Press, Cambridge, 2003).
- ⁴⁹ P. Brumer and M. Shapiro, *Adv. Chem. Phys.* **70**, 365 (1988).
- ⁵⁰ S. A. Rice, *Adv. Chem. Phys.* **XLVII**, 117 (1981).
- ⁵¹ N. De Leon and B. J. Berne, *J. Chem. Phys.* **75**, 3495 (1981).
- ⁵² S. A. Rice and M. S. Zhao, *Int. J. Quantum Chem.* **58**, 593 (1996).
- ⁵³ B. K. Carpenter, *Annu. Rev. Phys. Chem.* **56**, 57 (2005).
- ⁵⁴ G. S. Ezra, H. Waalkens, and S. Wiggins, *J. Chem. Phys.* **130**, 164118 (2009).
- ⁵⁵ S. Wiggins, *Physica D* **44**, 471 (1990).
- ⁵⁶ S. Wiggins, *Chaotic Transport in Dynamical Systems* (Springer-Verlag, New York, 1992).
- ⁵⁷ S. Wiggins, *Normally Hyperbolic Invariant Manifolds in Dynamical Systems* (Springer-Verlag, New York, 1994).
- ⁵⁸ A. M. O. de Almeida, N. De Leon, M. A. Mehta, and C. C. Marston, *Physica D* **46**, 265 (1990).
- ⁵⁹ N. De Leon, M. A. Mehta, and R. Q. Topper, *J. Chem. Phys.* **94**, 8310 (1991).
- ⁶⁰ N. De Leon, M. A. Mehta, and R. Q. Topper, *J. Chem. Phys.* **94**, 8329 (1991).
- ⁶¹ S. Wiggins, L. Wiesenfeld, C. Jaffe, and T. Uzer, *Phys. Rev. Lett.* **86**, 5478 (2001).
- ⁶² T. Komatsuzaki and R. S. Berry, *Adv. Chem. Phys.* **123**, 79 (2002).
- ⁶³ T. Komatsuzaki, K. Hoshino, and Y. Matsunaga, *Adv. Chem. Phys.*-Part B **130**, 257 (2005).
- ⁶⁴ C. Jaffé, K. Shinnosuke, J. Palacian, P. Yanguas, and T. Uzer, *Adv. Chem. Phys.*-Part B **130**, 171 (2005).
- ⁶⁵ L. Wiesenfeld, *Adv. Chem. Phys.*-Part B **130**, 217 (2005).
- ⁶⁶ H. Waalkens, R. Schubert, and S. Wiggins, *Nonlinearity* **21**, R1 (2008).
- ⁶⁷ P. Pechukas, *Annu. Rev. Phys. Chem.* **32**, 159 (1981).
- ⁶⁸ U. Lourderaj and W. L. Hase, *J. Phys. Chem. A* **113**, 2236 (2009).
- ⁶⁹ H. A. Posch, W. G. Hoover, and F. J. Vesely, *Phys. Rev. A* **33**, 4253 (1986).
- ⁷⁰ H. A. Posch and W. G. Hoover, *Phys. Rev. E* **55**, 6803 (1997).
- ⁷¹ W. G. Hoover, *J. Chem. Phys.* **109**, 4164 (1998).
- ⁷² W. G. Hoover, C. G. Hoover, and D. J. Isbister, *Phys. Rev. E* **63**, 026209 (2001).
- ⁷³ M. D'Alessandro, A. Tenenbaum, and A. Amadei, *J. Phys. Chem. B* **106**, 5050 (2002).
- ⁷⁴ P. Minary, G. J. Martyna, and M. E. Tuckerman, *J. Chem. Phys.* **118**, 2510 (2003).
- ⁷⁵ P. Minary, G. J. Martyna, and M. E. Tuckerman, *J. Chem. Phys.* **118**, 2527 (2003).
- ⁷⁶ P. Minary and M. E. Tuckerman, *J. Am. Chem. Soc.* **126**, 13920 (2004).
- ⁷⁷ P. Minary and M. E. Tuckerman, *J. Am. Chem. Soc.* **127**, 1110 (2005).
- ⁷⁸ M. Litniewski, *J. Phys. Chem.* **97**, 3842 (1993).
- ⁷⁹ T. Morishita, *J. Chem. Phys.* **119**, 7075 (2003).
- ⁸⁰ G. S. Ezra and S. Wiggins, *J. Phys. A* **42**, 042001 (2009).
- ⁸¹ N. B. Slater, *J. Chem. Phys.* **24**, 1256 (1956).
- ⁸² E. Thiele, *J. Chem. Phys.* **36**, 1466 (1962).
- ⁸³ R. S. Dumont and P. Brumer, *J. Phys. Chem.* **90**, 3509 (1986).
- ⁸⁴ P. Brumer, D. E. Fitz, and D. Wardlaw, *J. Chem. Phys.* **72**, 386 (1980).
- ⁸⁵ E. Pollak, *J. Chem. Phys.* **74**, 6763 (1981).
- ⁸⁶ J. Binney, O. E. Gerhard, and P. Hut, *Mon. Not. R. Astron. Soc.* **215**, 59 (1985).
- ⁸⁷ H.-D. Meyer, *J. Chem. Phys.* **84**, 3147 (1986).
- ⁸⁸ H. Waalkens, A. Burbanks, and S. Wiggins, *Phys. Rev. Lett.* **95**, 084301 (2005).
- ⁸⁹ H. Waalkens, A. Burbanks, and S. Wiggins, *J. Phys. A* **38**, L759 (2005).
- ⁹⁰ D. J. Evans, W. G. Hoover, B. H. Failor, B. Moran, and A. J. C. Ladd, *Phys. Rev. A* **28**, 1016 (1983).
- ⁹¹ D. J. Evans and G. P. Morriss, *Phys. Lett. A* **98**, 433 (1983).
- ⁹² V. Szebehely, *Theory of Orbits* (Academic, New York, 1967).
- ⁹³ L. Verlet, *Phys. Rev.* **159**, 98 (1967).
- ⁹⁴ J. R. Silvester, *Math. Gaz.* **84**, 460 (2000).
- ⁹⁵ E. P. Wigner, *J. Chem. Phys.* **7**, 646 (1939).
- ⁹⁶ J. C. Keck, *Adv. Chem. Phys.* **XIII**, 85 (1967).
- ⁹⁷ J. B. Anderson, *Adv. Chem. Phys.* **XCI**, 381 (1995).
- ⁹⁸ T. Uzer, C. Jaffe, J. Palacian, P. Yanguas, and S. Wiggins, *Nonlinearity* **15**, 957 (2002).
- ⁹⁹ H. Waalkens, A. Burbanks, and S. Wiggins, *J. Phys. A* **37**, L257 (2004).
- ¹⁰⁰ H. Waalkens and S. Wiggins, *J. Phys. A* **37**, L435 (2004).
- ¹⁰¹ H. Waalkens, A. Burbanks, and S. Wiggins, *J. Chem. Phys.* **121**, 6207 (2004).
- ¹⁰² R. Schubert, H. Waalkens, and S. Wiggins, *Phys. Rev. Lett.* **96**, 218302 (2006).
- ¹⁰³ T. Komatsuzaki and R. S. Berry, *J. Mol. Struct.: THEOCHEM* **506**, 55 (2000).
- ¹⁰⁴ R. S. Dumont, *J. Chem. Phys.* **91**, 4679 (1989).
- ¹⁰⁵ R. S. Dumont, *J. Chem. Phys.* **91**, 6839 (1989).
- ¹⁰⁶ D. Chandler, *J. Chem. Phys.* **68**, 2959 (1978).
- ¹⁰⁷ D. Chandler, *Introduction to Modern Statistical Mechanics* (Oxford University Press, New York, 1987).
- ¹⁰⁸ S. K. Gray and S. A. Rice, *J. Chem. Phys.* **86**, 2020 (1987).
- ¹⁰⁹ E. Pollak and P. Pechukas, *J. Chem. Phys.* **69**, 1218 (1978).
- ¹¹⁰ D. G. Truhlar, B. C. Garrett, and S. J. Klippenstein, *J. Phys. Chem.* **100**, 12771 (1996).
- ¹¹¹ Such a separation is assumed to be meaningful for the range of energies considered here.
- ¹¹² B. J. Berne, N. DeLeon, and R. O. Rosenberg, *J. Phys. Chem.* **86**, 2166 (1982).
- ¹¹³ V. I. Arnold, *Mathematical Methods of Classical Mechanics* (Springer-Verlag, New York, 1978).
- ¹¹⁴ M. Toller, G. Jacucci, G. DeLorenzi, and C. P. Flynn, *Phys. Rev. B* **32**, 2082 (1985).
- ¹¹⁵ R. S. Mackay, *Phys. Lett. A* **145**, 425 (1990).
- ¹¹⁶ R. E. Gillilan, *J. Chem. Phys.* **93**, 5300 (1990).
- ¹¹⁷ W. L. Hase, D. G. Buckowski, and K. N. Swamy, *J. Phys. Chem.* **87**, 2754 (1983).
- ¹¹⁸ M. Berblinger and C. Schlier, *J. Chem. Phys.* **101**, 4750 (1994).
- ¹¹⁹ S. Y. Grebenshchikov, R. Schinke, and W. L. Hase, in *Unimolecular Kinetics: Part 1. The Reaction Step*, Comprehensive Chemical Kinetics Vol. 39, edited by N. J. B. Greene (Elsevier, New York, 2003), pp. 105–242.
- ¹²⁰ J. N. Stember and G. S. Ezra, *Chem. Phys.* **337**, 11 (2007).
- ¹²¹ N. B. Slater, *Theory of Unimolecular Reactions* (Cornell University Press, Ithaca, NY, 1959).
- ¹²² D. L. Bunker, *J. Chem. Phys.* **37**, 393 (1962).
- ¹²³ D. L. Bunker, *J. Chem. Phys.* **40**, 1946 (1964).
- ¹²⁴ D. L. Bunker and W. L. Hase, *J. Chem. Phys.* **59**, 4621 (1973).
- ¹²⁵ E. Thiele, *J. Chem. Phys.* **38**, 1959 (1963).
- ¹²⁶ D. L. Bunker, *Theory of Elementary Gas Reaction Rates* (Pergamon, Oxford, 1966).
- ¹²⁷ D. L. Bunker and M. L. Pattengil, *J. Chem. Phys.* **48**, 772 (1968).
- ¹²⁸ W. L. Hase, in *Modern Theoretical Chemistry*, edited by W. H. Miller (Plenum, New York, 1976), Vol. 2, pp. 121–170.
- ¹²⁹ S. F. Chekmarev, *Phys. Rev. E* **78**, 066113 (2008).
- ¹³⁰ Wolfram Research, Inc., *MATHEMATICA v7, Champaign, IL, 2008*.
- ¹³¹ M. Lutz, *Programming Python*, 3rd ed. (O'Reilly Media, New York, 2006).